EFFECTS OF DIFFERENT BATTER FORMULATIONS ON QUALITY OF DEEP-FAT FRIED CARROT SLICES

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ABSTRACT

EFFECTS OF DIFFERENT BATTER FORMULATIONS ON QUALITY OF DEEP-FAT FRIED CARROT SLICES

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The main objective of the study was to evaluate the effects of starch and gum types on product quality of deep-fat fried carrot slices. It was also aimed to evaluate the applicability of image processing for determination of oil uptake.

In the first part of the study, carrot slices were dipped into batters containing three different concentrations of dextrin or pre-gelatinized tapioca starch and fried for 2, 3, and 4 minutes at 170±2 °C. Coating pick-up of batter formulations and moisture content, oil content, frying yield, bulk density, porosity, texture and color of fried slices were evaluated. In the second part of the study, the effects of different gum types (HPMC, xanthan gum, guar gum, guar-xanthan gum combination) on quality attributes were studied. No starch or gum added coating formulation was used as the control. Finally, images of carrot and batter sections of the fried samples were obtained using digital camera and area fractions of oil droplets were determined using image processing.

Acceptable product quality was obtained at higher concentrations of pregelatinized tapioca starch. On the other hand, increasing dextrin concentrations had an adverse affect on the product quality.

As a result of the study, guar-xanthan gum combination has been found as the most effective additive on the batter performance. This additive provided the highest moisture content, lowest oil content, highest volume and lightest color to the product after frying. The porous and crunchy structure obtained using this combination was at the acceptable level for deep-fat fried products.

Determination of area fraction of oil droplets on carrot and batter surfaces of fried samples using image processing was correlated with the oil content of fried product at initial stages of frying.

The correlation coefficient between moisture content and frying yield was found as 0.90. A correlation was also determined between oil content and moisture content (r= -0.88).

Keywords: Batter, Carrot, Frying, Gums, Physical properties, Starches.

DEĞİŞİK KAPLAMA FORMÜLASYONLARININ KIZARTILMIŞ HAVUÇ DİLİMLERİNİN KALİTESİ ÜZERİNE ETKİSİ

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Bu çalışmanın temel amacı, nişasta ve sakız çeşitlerinin kızartılmış havuç dilimlerinin kalitesi üzerine etkisinin değerlendirilmesidir. Ayrıca görüntü işleminin yağ emiliminin belirlenmesinde uygulanabilirliğinin tespiti amaçlanmaktadır.

İlk bölümde, havuç dilimleri önceden jelatinize edilmiş tapioka nişastası veya dekstrinin üç farklı konsantrasyonunu (1%, 3%, 5%) içeren hamurlara batırılmış ve 170±2°C'de 2, 3, 4 dakika kızartılmıştır. Kızarmış dilimlerin kaplama tutması, nem miktarı, yağ miktarı, kızartma verimi, yoğunluğu, gözenekliliği, tekstürü ve rengi değerlendirilmiştir. İkinci kısımda, değişik zamk çeşitlerinin (HPMC, guar zamkı, ksantan zamkı ve guar-ksantan zamk kombinasyonunun kalite niteliklerine olan etkileri çalışılmıştır. Nişasta veya zamk eklenmemiş kaplama formülasyon

kontrol olarak kullanılmıştır. Son olarak, kızartılmış numunenin havuç ve kaplama kısımlarının görüntüleri dijital kamera ile elde edilmiş ve yağ damlacıklarının kapladıkları alan fraksiyonları görüntü işlemi ile tespit edilmiştir.

Nişastanın nispeten yüksek oranlarda kullanımı arzu edilen ürünlerin elde edilmesini sağlamıştır. Diğer yandan, dekstrin konsantrasyonundaki artışın ürün kalitesi üzerine olumsuz etkide bulunduğu gözlenmiştir.

Çalışmanın sonucunda, guar-ksantan zamk kombinasyonu hamur performansında en etkili katkı olarak bulunmuştur. Bu katkı ürüne kızartma sonrasında en yüksek nem oranını, en düşük yağ içeriğini, en büyük hacmi ve en açık rengi sağlamıştır. Bu kombinasyonun kullanımı sırasında elde edilen gözenekli ve çıtır yapı derin yağda kızartılmış ürünler için kabul edilebilir düzeydedir.

Kızartılmış numunelerin havuç ve kaplama kısımlarındaki yağ damlacıklarının kapladıkları alanın fraksiyonunun görüntü işlemi ile tespiti kızartmanın başlangıç aşamasındaki yağ içeriği hakkında fikir vermektedir.

Nem içeriğiyle kızartma verimi arasındaki korelasyon katsayısı 0.90 olarak bulunmuştur. Yağ içeriğiyle nem içeriği arasında da korelasyon tespit edilmiştir (r=-0.88).

Anahtar sözcükler: Havuç, Fiziksel özellikler, Kaplama hamuru, Kızartma, Nişasta, Zamk.

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TABLE OF CONTENTS

ABSTRACT	iv
ÖZ	vi
ACKNOWLEDGMENTS	viii
TABLE OF CONTENTS	ix
LIST OF TABLES	xii
LIST OF FIGURES	XV

CHAPTER

1.	INR	ODUCT	'ION		1
	1.1	Carrot			1
	1.2	Deep-F	at Frying I	Mechanism	1
	1.3	Quality	Parameter	rs in Fried Foods	3
		1.3.1	Oil and I	Moisture Contents	3
		1.3.2	Texture.		6
		1.3.3	Color		7
		1.3.4	Porosity		8
	1.4	Batter S	Systems		9
		1.4.1	Starches		11
		1.4.2	Proteins		14
		1.4.3	Gums		14
			1.4.3.1	Cellulose Derivatives	15
			1.4.3.2	Xanthan Gum	16
			1.4.3.3	Guar Gum	16
			1.4.3.4	Synergistic Interactions	17
	1.5	Objecti	ves of the	Study	17
2.	MA	TERIAL	S AND M	ETHODS	19

2.1	Material	ls	19	
2.2	Batter P	reparation	20	
2.3	Sample	Preparation and Frying Conditions	20	
2.4	.4 Analysis of Sample			
	2.4.1	Coating Pick-Up Calculations	21	
	2.4.2	Moisture Analysis	21	
	2.4.3	Oil Analysis	21	
	2.4.4	Frying Yield Calculations	22	
	2.4.5	Colorimetric Measurements	22	
	2.4.6	True and Bulk Volume Measurements	23	
	2.4.7	True and Bulk Density Calculations	24	
	2.4.8	Porosity Calculations	24	
	2.4.9	Textural Measurements	24	
	2.4.10	Image Processing	25	
	2.4.11	Statistical Analysis	25	
RES	ULTS AI	ND DISCUSSION	26	
3.1	Effects Paramet	of Tapioca Starch and Dextrin on the Quality ers of Deep-fat Fried Carrots Slices	26	
	3.1.1	Coating Pick-Up	26	
	312		20	
	J.1.4	Moisture Content	28	
	3.1.3	Moisture Content Oil Content	28 29	
	3.1.3 3.1.4	Moisture Content Oil Content Frying Yield	28 29 31	
	3.1.2 3.1.3 3.1.4 3.1.5	Moisture Content Oil Content Frying Yield Crispness	28 29 31 32	
	3.1.2 3.1.3 3.1.4 3.1.5 3.1.6	Moisture Content Oil Content Frying Yield Crispness Bulk Density	28 29 31 32 34	
	3.1.2 3.1.3 3.1.4 3.1.5 3.1.6 3.1.7	Moisture Content Oil Content Frying Yield Crispness Bulk Density Porosity	28 29 31 32 34 36	
	3.1.2 3.1.3 3.1.4 3.1.5 3.1.6 3.1.7 3.1.8	Moisture Content Oil Content Frying Yield Crispness Bulk Density Porosity Color	28 29 31 32 34 36 37	
3.2	3.1.2 3.1.3 3.1.4 3.1.5 3.1.6 3.1.7 3.1.8 Effects Deep-Fa	Moisture Content Oil Content Frying Yield Crispness Bulk Density Porosity Color of Different Gum Types on Quality Parameters of it Fried Carrot Slices	28 29 31 32 34 36 37 41	
3.2	3.1.2 3.1.3 3.1.4 3.1.5 3.1.6 3.1.7 3.1.8 Effects Deep-Fa 3.2.1	Moisture Content Oil Content Frying Yield Crispness Bulk Density Porosity Color of Different Gum Types on Quality Parameters of tt Fried Carrot Slices	28 29 31 32 34 36 37 41 41	
3.2	3.1.2 3.1.3 3.1.4 3.1.5 3.1.6 3.1.7 3.1.8 Effects Deep-Fa 3.2.1 3.2.2	Moisture Content Oil Content Frying Yield Crispness Bulk Density Porosity Color of Different Gum Types on Quality Parameters of th Fried Carrot Slices Moisture content	28 29 31 32 34 36 37 41 41 41	

3.

		3.2.3	Oil Content	44
		3.2.4	Frying Yield	45
		3.2.5	Crispness	46
		3.2.6	Bulk Density	47
		3.2.7	Porosity	49
		3.2.8	Color	50
	3.3	Compar	ison of the Effects of Tapioca Starch, Dextrin and Gum	
		Types of	n Deep-Fat Fried Carrot Slices	54
	3.4	Image A	nalysis	61
4.	CON	ICLUSIC	N AND RECOMMENDATIONS	64
REFERE	ENCE	S		66
APPENI	DICES	5		
A.	TEX	TURE PI	ROFILE ANALYSIS	76
B.	ANC	OVA ANI	D DUNCAN TABLES	77
C.	WA	TER BIN	DING CAPACITY	104
D.	FIG	URES OF	DEEP-FAT FRIED CARROT SLICES	105
E.	SUR	FACE PI	LOTS OF IMAGE PROCESSING	108

LIST OF TABLES

TABLE		
1.1	Typical factors affecting oil uptake	5
1.2	Concentrations and functionality of ingredients used in batter formulations	10
1.3	Amylose and amylopectin content of starches	13
3.1	Area fraction and average size of droplets observed on surfaces of carrot and batter portions of the fried sample	62
B.1	ANOVA and Duncan's Multiple Range Test Table for coating pick- up of fried samples with different concentrations of dextrin and pre- gelatinized tapioca starch	77
B.2	ANOVA and Duncan's Multiple Range Test Table for moisture content of fried samples with different concentrations of dextrin and pre-gelatinized tapioca starch.	78
B.3	ANOVA and Duncan's Multiple Range Test Table for oil content of fried samples with different concentrations of dextrin and pre- gelatinized tapioca starch	79
B.4	ANOVA and Duncan's Multiple Range Test Table for frying yield of samples with different concentrations of dextrin and pre-gelatinized tapioca starch	80
B.5	ANOVA and Duncan's Multiple Range Test Table for fracturability of fried samples with different concentrations of dextrin and pre- gelatinized tapioca starch.	81
B.6	ANOVA and Duncan's Multiple Range Test Table for bulk density of fried samples with different concentrations of dextrin and pre- gelatinized tapioca starch	82

B.7	ANOVA and Duncan's Multiple Range Test Table for porosity of fried samples with different concentrations of dextrin and pre- gelatinized tapioca starch.	83
B.8	ANOVA and Duncan's Multiple Range Test Table for Hunter L value of fried samples with different concentrations of dextrin and pre-gelatinized tapioca starch	84
B.9	ANOVA and Duncan's Multiple Range Test Table for Hunter a value of fried samples with different concentrations of dextrin and pre-gelatinized tapioca starch	85
B.10	ANOVA and Duncan's Multiple Range Test Table for ΔE of fried samples with different concentrations of dextrin and pre-gelatinized tapioca starch.	86
B.11	ANOVA and Duncan's Multiple Range Test Table for coating pick- up of fried samples with different gum types	87
B.12	ANOVA and Duncan's Multiple Range Test Table for moisture content of fried samples with different gum types	88
B.13	ANOVA and Duncan's Multiple Range Test Table for oil content of fried samples with different gum types	89
B.14	ANOVA and Duncan's Multiple Range Test Table for frying yield of samples with different gum types	90
B.15	ANOVA and Duncan's Multiple Range Test Table for fracturability of fried samples with different gum types	91
B.16	ANOVA and Duncan's Multiple Range Test Table for bulk density of fried samples with different gum types	92
B.17	ANOVA and Duncan's Multiple Range Test Table for porosity of fried samples with different gum types	93
B.18	ANOVA and Duncan's Multiple Range Test Table for Hunter L value of fried samples with different gum types	94
B.19	ANOVA and Duncan's Multiple Range Test Table for Hunter a value of fried samples with different gum types	95
B.20	ANOVA and Duncan's Multiple Range Test Table for ΔE of fried samples with different starch or gum types	96

B.21	ANOVA and Duncan's Multiple Range Test Table for coating pick- up of fried samples with different starch or gum types	97
B.22	ANOVA and Duncan's Multiple Range Test Table for moisture content of fried samples with different starch or gum types.	98
B.23	ANOVA and Duncan's Multiple Range Test Table for oil content of fried samples with different starch or gum types	99
B.24	ANOVA and Duncan's Multiple Range Test Table for fracturability	
	of fried samples with different starch or gum types	100
B.25	ANOVA and Duncan's Multiple Range Test Table for bulk density of fried samples with different starch or gum types	101
B.26	ANOVA and Duncan's Multiple Range Test Table for Hunter L value of fried samples with different starch or gum types	102
B.27	ANOVA and Duncan's Multiple Range Test Table for Hunter a value of fried samples with different starch or gum types	103
C.1	Water binding capacities (WBC) of different starches and gums	104

LIST OF FIGURES

FIGURE

3.1	Effects of pre-gelatinized tapioca starch and dextrin at different concentrations on coating pick-up of deep-fat fried carrot slices	27
3.2	Effects of pre-gelatinized tapioca starch and dextrin at different concentrations on moisture content of deep-fat fried carrot slices	29
3.3	Effects of pre-gelatinized tapioca starch and dextrin at different concentrations on oil uptake of deep-fat fried carrot slices	30
3.4	Effects of pre-gelatinized tapioca starch and dextrin at different concentrations on frying yield of deep-fat fried carrot slices	32
3.5	Effects of pre-gelatinized tapioca starch and dextrin at different concentrations on fracturability of deep-fat fried carrot slices	34
3.6	Effects of pre-gelatinized tapioca starch and dextrin at different concentrations on bulk density of deep-fat fried carrot slices	35
3.7	Effects of pre-gelatinized tapioca starch and dextrin at different concentrations on porosity of deep-fat fried carrot slices	36
3.8	Effects of pre-gelatinized tapioca starch and dextrin at different concentrations on Hunter L value of deep-fat fried carrot slices	37
3.9	Effects of pre-gelatinized tapioca starch and dextrin at different concentrations on Hunter a value of deep-fat fried carrot slices	38
3.10	Effects of pre-gelatinized tapioca starch and dextrin at different concentrations on Hunter b value of deep-fat fried carrot slices	39
3.11	Effects of pre-gelatinized tapioca starch and dextrin at different concentrations on ΔE of deep-fat fried carrot slices	40
3.12	Effects of gum types on coating pick-up of deep-fat fried carrot slices	42
3.13	Effects of gum types on moisture content of deep-fat fried carrot	
	slices	43

3.14	Effects of gum types on oil content of deep-fat fried carrot slices	44
3.15	Effects of gum types on frying yield of deep-fat fried carrot slices	46
3.16	Effects of gum types on fracturability of deep-fat fried carrot slices	47
3.17	Effects of gum types on bulk density of deep-fat fried carrot slices	48
3.18	Effects of gum types on porosity of deep-fat fried carrot slices	50
3.19	Effects of gum types on Hunter L value of deep-fat fried carrot slices.	51
3.20	Effects of gum types on Hunter a value of deep-fat fried carrot slices.	52
3.21	Effects of gum types on Hunter b value of deep-fat fried carrot slices.	53
3.22	Effects of gum types on ΔE value of deep-fat fried carrot slices	54
3.23	Effects of different hydrocolloids on coating pick-up of deep-fat fried carrot slices	55
3.24	Effects of different hydrocolloids on moisture content of deep-fat fried carrot slices.	56
3.25	Effects of different hydrocolloids on oil content of deep-fat fried carrot slices.	57
3.26	Effects of different hydrocolloids on fracturability of deep-fat fried carrot slices	58
3.27	Effects of different hydrocolloids on bulk density of deep-fat fried carrot slices.	59
3.28	Effects of different hydrocolloids on Hunter L value of deep-fat fried carrot slices	60
3.29	Effects of different hydrocolloids on Hunter a value of deep-fat fried carrot slices.	61
A.1	Typical TPA curve (guar-xanthan gum combination coated carrot slices; 3 minutes fried.)	76
D.1	Carrot and control batter fried for 2 & 4 minutes	105
D.2	Carrot and 1% dextrin added batter fried for 2 & 4 minutes	105

D.3	Carrot and 5% dextrin added batter fried for 2 & 4 minutes	105
D.4	Carrot and 1% pre-gelatinized tapioca starch added batter fried for 2 & 4 minutes.	106
D.5	Carrot and 5% pre-gelatinized tapioca starch added batter fried for 2 & 4 minutes	106
D.6	Carrot and xanthan gum added batter fried for 2 & 4 minutes	106
D.7	Carrot and HPMC added batter fried for 2 & 4 minutes	107
D.8	Carrot and guar-xanthan gum combination added batter fried for 2 & 4 minutes	107
E.1	Surface plot of control coated carrot	108
E.2	Surface plot of for guar-xanthan gum combination added batter coated carrot	109

CHAPTER 1

INTRODUCTION

1.1 Carrot

Yellow types carrots were selected and cultured in Syria and Turkey in the 9 or 10th century; then spread to China in the late 13th century and to Europe in the 14th century (www.uga.edu/vegetable/carrot.html).

Today, carrot (*Daucus carota*) is one of the most popular vegetables consumed in the world whatever the season is. In addition, consumption of carrot has been increasing regularly, particularly for cooked carrots.

Carrots have moisture content of 80-90% (wb) at the time of harvest and they are highly susceptible to moisture loss. Carrot is one of the root vegetable rich in fibers and carotenoids, which are associated respectively to cholesterol metabolism and antioxidant protection (Nicolle et al., 2003). It contains 12 % dietary fiber, 5 % sugar, 1 % protein and 5.6 mg carotenoids/100 g besides being rich in B vitamins, vitamin C, potassium, sodium and magnesium (http://www.nutritiondata.com).

1.2 Deep-Fat Frying Mechanism

Deep-fat frying is a dry cooking process, which consists basically the immersion of food slices in hot vegetable oil (Moyano et al., 2002). It is a widely used method for preparing tasty foods that have soft and moist interior together with the crispy crust (Garcia et al., 2001). Throughout frying process, physical, chemical and sensory characteristics of the food are modified. The main purpose

of deep-fat frying is to retain of all the flavors and juices inside a crisp crust by immersing the food in the hot oil (Moyano et al., 2002).

During frying, simultaneous heat and mass transfer occur. Upon addition of the food to the hot oil, the surface temperature of the food rises and the water at the surface immediately starts boiling. Due to the evaporation, surface drying is seen. The evaporation also leads to shrinkage and crust formation (Mellema, 2003). Heat transferred from the oil to the food causes conversion of inner moisture to steam, which creates a pressure gradient as the surface dries out. By the help of capillaries and channels in the cellular structure this pressure gradient within the product gently 'pumps' the water from the core of the food to the crust, which will remove during frying. At the same time, oil adheres to product's surface at the damaged areas and enters the voids left by the water vapor (Debnath et al., 2003). The fact that the vapor leaves voids for the oil to enter later, is the reason why the moisture content of the food largely determines oil uptake (Gamble et al., 1987a; Lamberg et al., 1990; Mehta and Swinburn, 2001;Saguy and Pinthus, 1995; Southern et al, 2000).

Especially at high moisture content, vapor protects the food from oil absorption by creating an overpressure inside the pores. This barrier property of vapor probably continues until a few seconds after removal of the food from the oil. After taking the food out of the fryer the temperature drops and the vapor in the pores condensates (Mellema, 2003). This condensation mechanism creates vacuum effect, which causes the adhering oil being pulled into the product (Debnath et al., 2003). For tortilla chips while 80% of the oil remains at the surface of the product only 20% is present internally at the end of frying. During cooling about 64% of this surface oil is later absorbed into the interior (Moreira et al., 1997).

Water plays a number of roles during frying process. Firstly, it takes away thermal energy from the hot frying oil surrounding the frying food. This removal of energy by conversion of liquid water to steam prevents burning caused by excessive dehydration at the surface of the product. Therefore, despite the fact that the oil may be at 170°C, the frying food is only about 100°C, which represents phase change temperature. Another function of water is to cook the interior of frying food. As known water is a better conductor than the fat, protein, and carbohydrate portions of food, which facilitates to conduct heat energy from the surfaces contacted by hot frying oil to the interior. On the other hand, migration of water from the central portion radially outward to the walls and edges causes to the movement of water-soluble materials to the exterior and leaching liquified food components from the food (Blumenthal, 1991).

1.3 Quality Parameters in Fried Foods

In general, the four principal quality factors in foods are: (1) appearance, including color, shape, etc.; (2) flavor, including taste and odor; (3) texture; and (4) nutrition (Bourne, 1982).

In fried foods the most important product properties that are measured to determine related quality characteristics are discussed in this part of the study. These properties are: moisture content, oil content, color, texture and porosity.

1.3.1 Oil and Moisture Contents

Oil content is one of the most important quality attributes of a deep fat fried product. The texture of a low-oil-content product can be hard and unpleasant. However, the high oil content is costly to the processor and results in an oily and tasteless product (Moreira et al, 1999). Also, with the growing healthy consciousness of the consumer, demand for lower oil-content fried foods has increased. Therefore, oil contents of products have to be taken into consideration.

Foods with more moisture loss also show more oil uptake (Gamble et al, 1987b). Some even argue that the total volume of oil will equal the total volume of water removed (Pinthus et al., 1993).

Oil uptake during deep-fat frying of products is affected by many factors, including oil quality, frying temperature and duration, its composition (e.g. moisture, solids), porosity, pre-frying treatments (e.g., drying, blanching) and coating (Pinthus et al., 1995b; Selman and Hopkins, 1989; Stier and Blumenthal, 1990). All these factors affecting oil uptake of product during frying are summarized in Table 1.1.

Since most of the fat is taken up after removal of the food from the oil so, the habits of the consumer during removal of the food from the oil can play large role. Proper shaking and draining of the food are important for reducing oil content of the food (Mellema, 2003).

Excess oil absorption may result from low frying temperatures or overloading the fryer beyond its capacity. At low temperatures, there is a tendency to cook food longer to obtain the desired color of the food. Therefore, oil absorption increases (Orthoefer et al., 1996). In contrast, Moreira et al. (1999) argued that higher oil temperatures lead to a faster crust formation and so favoring the conditions for oil absorption.

It is well known that oil uptake is a function of the surface area of the food, thus it is obvious that the shape of the food will affect total oil uptake. For instance, samples can be sliced in larger chunks or surface roughness can be reduced by control of the quality of the slicing blades (Mellema, 2003).

As previously mentioned, one of the most often mentioned parameters to reduce oil uptake at the level of the food composition is the moisture content. Predrying of foods like potatoes is a common way to reduce oil uptake (Krokida et al., 2001).

Since the properties of the surface of the food are most important for oil uptake, the application of a coating is a promising route. Often mentioned properties of coatings in relation to oil uptake are low moisture content, low moisture permeability, thermo-gelling or cross-linked (Mellema, 2003).

Tabl	e 1.1	Typical	l factors	affecting	oil u	iptake
		2		0		

Factor	Reference
Increasing	
Surface roughness	Rubnov and Saguy (1994)
Thinner product	Krokida et al. (2000)
Increased area	Keller et al. (1990)
Porosity	Pinthus et al. (1995b)
<u>Decreasing</u>	
Pre-drying	Krokida et al. (2001)
Lower initial moisture content	Krokida et al. (2000)
Coating	Khalil (1999); Rayner et al. (2000); Shih et al. (2001)

There are abundant methods to determine oil content of products. Soxhlet extraction is a simple gravimetric method, in which the oil is extracted from the product using organic solvents. In DSC method, the melting enthalpy is taken as a measure of oil. MRI (magnetic Resonance imaging) method relies on the difference in relaxation between solids and liquids (Mellema, 2003). Ufheil and Escher (1996) followed the uptake of oil during deep-fat frying of potato slices by frying slices for an equal length of time, introducing oil soluble and heat stable dye into the oil at different times. Gamble et al. (1987b) investigated the distribution of

oil taken up during frying. Samples were fried in red-stained oil and after frying products were photographed.

1.3.2 Texture

The term texture is still not well defined in food technology; but it is a very important quality characteristic of the fried product. An important texture characteristic for fried products is crispness without being very hard. Crispness indicates freshness and high quality (Szczesniak, 1988). A lack of crispness can be defined as either a chewy toughness or a mushy softness (Fizsman and Salvador, 2003).

The crispness is a phenomenon with two components: oral and aural. The tactile sensation of the teeth biting through the food and the sound produced inside the head as the teeth cause the crushing and collapse of a multitude of small cells within the product. Ideally, the crust should exhibit a structure that sufficiently resists the initial bite but then disappears quickly in the mouth (Loewe, 1993).

The crisp final texture of the fried product can be investigated by means of instrumental or sensory techniques. Parameters such as crispness or crunchiness, fragility, tenderness, etc., are hard to quantify using instrumental techniques because what is perceived in the mouth is a complex of sensations (Fizsman and Salvador, 2003).

Puncture with a plunger is the most used technique for the measurements of texture parameters (Fan et al., 1997; Mohammed et al., 1998). Other parameters such as greasiness, juiciness, oiliness and mealiness of products can be assessed with trained panelist (Prakash and Rajalakshimi, 1999).

The overall texture of a fried product is partially influenced by the composition of a food material. Interactions between proteins, starch, and its

components (amylose and amylopectin) are of importance for the final quality of the product (Rovedo et al., 1999).

Olewnick and Kulp (1993) studied how some characteristics of wheat flour affect the behavior of tempura type-batters during frying, when applied on chicken drumsticks. In this study, the crispness of products was evaluated organoleptically. Salvador et al. (2002) tried effects of corn flour, salt and leavening on the texture of fried, battered squid rings. It was reported that the ingredient having the greatest effect on the final texture of the coating layer of the fried product is the leavening agent. Leavening agent contributes to the crispness and tenderness required for the fried outer crust of this product type.

1.3.3 Color

Color is an important factor influencing consumer acceptability of a fried product. It can indicate high-quality products such as the golden yellow of a potato. Color also influences flavor recognition. Panel evaluation and comparison to standards are the most common approaches for determining color of fried foods. Colorimeters can also be used to determine the color of products objectively. Hunter L, a and b color scale to express color differences among samples is commonly used. The L dimension defines the lightness, the a refers to the redness or greenness and the b dimension refers to the blueness or yellowness.

The consumer generally uses the color of a product in order to determine the end of the frying process. The final color of the fried product depends on the absorption of oil and the chemical reactions of browning of reducing sugars and protein sources (Baixauli et al., 2002). Caramelization, involving thermal degradation of sugars without amine participation also takes place during frying process (Baik and Mittal, 2003). Frying temperature and duration are directly effective on color development. Ling et al. (1998) found onion rings fried at 190°C had lower L values (decreased lightness), higher a values (increased redness) and lower b values (decreased yellowness) than onion rings fried at 170°C. Furthermore, similar color changes for coated chicken parts with increasing frying times were reported (Waimaleongora-Ek and Chen, 1983). Fried foods are also affected by the type and age of the frying oil (Loewe, 1990). Lee and Dawson (1973) showed that the adsorption of reused corn oil by chicken pieces would undoubtedly affect product quality.

Applied coating formulation and the colorant that may be included in it are effective in determining color of the final product. Corn flour and colorants (e.g.,riboflavin, tartrazine) added to the coating give a more yellow or orange color to the product, which otherwise would look pale (Baixauli et al., 2002).

Hanson and Fletcher (1963) studied the effects of flour type on color of deep-fat fried chicken parts. They reported wheat flour produced a grayish-brown color and yellow corn flour provided a greenish yellow color.

1.3.4 Porosity

Porosity is generally used for leavened batters and describes the open cellular network found in products. The porosity of the product formed during frying plays an important role in the oil uptake. When a crust begins to form at the surface of the sample, there is an excessive pressure buildup and the product expands and puffs. Low leavening level or low batter viscosity may cause decreased tempura puff (Loewe, 1990).

Frying process can change the product's porous structure by the phenomenon of shrinkage or puffing (Yamsaengsung and Moreira, 2002). Moreira, et al. (1995) reported that bulk density decreases and porosity and oil uptake increase with frying time during frying of tortilla chips. Llorca et al. (2001) showed that the CO_2 that forms during frying process because of the leavening

agent in the batter formulation is responsible, together with the released water vapor, for producing the pores and channels.

1.4 Batter Systems

The properties of the surface of the food are very important for oil uptake so the application of a coating is a promising route. Coatings can be thin and invisible or thick like a batter. The main difference is that batters may be more easily applied by the consumer and also they have less of the puncturing problems associated with thin coatings (Mellema, 2003).

A batter can be defined as liquid dough, basically consisting of flour and water, into which a product is dipped before frying whereas breading is a dry mixture and applied to the moistened or battered foods prior to cooking. Batter systems are classified into two categories: interface/adhesion and puff/tempura. The interface/adhesion batters are typically used with a supplement breading or breadcrumb. The batter serves, as an adhesive layer between the food surface and the breading and chemical leavening is not normally used. In puff/tempura batter systems both wheat and corn flours play an important role. They are chemically leavened and used as an outside coating for the food. The batter uniformity and thickness, which is related to the batter viscosity, determine acceptability of the finished product (Loewe, 1990).

The mode of action of batter in retarding oil absorption appears to be due to the rapid formation of a hard crust as a result of water loss, the crust being relatively impervious to the movement of water and oil (Love and Goodwin, 1974). The ability of batter to form a crust is enhanced by the higher initial amount of coating adhering.

Tempura-type batters form crisp and uniform layer over the food, constituting its final outer coating. Batters enhance the texture, flavor and appearance of foods. They act as a barrier against loss of moisture by protecting the natural juices of foods, thereby ensuring a final product that is tender and juicy on the inside and at the same time crisp on the outside (Fiszman and Salvador, 2003).

In practice, the list of ingredients in batter systems is quite long (starch, salt, leavening agent, gums and many other items) and batters have therefore become highly sophisticated, complex systems in which the nature of ingredients is very wide-ranging and their interaction determines the final performance of the product (Table 1.2). The most important ingredients were explained below in more detail.

Table 1.2 Concentrations and functionality of ingredients used in batter formulations (Fiszman and Salvador, 2003).

Ingredient	Addition range (%)	Functionality or effect on quality
Wheat flour	> 40	Body structure, viscosity
Corn flour	> 30	Crispness, golden brown color
Starches	≤ 5	Changes in tenderness and crunchiness
Leavening agents	< 3	Porous structure
Gums	≤ 1	Viscosity control, ability to participate in gel/film formation
Salt, sugars, dextrins	At different concentrations	Product quality improvement

1.4.1 Starches

The theoretical explanation of how wheat and corn flours affect the structure of batter coatings focuses upon the complementary actions of the protein and starch components.

Starch occurs widely in the nature and is the most commonly used food hydrocolloid. This is because of the wide range of functional properties it can provide in its natural and modified forms. It is a mixture of a linear polymer (amylose) and a branched-chain polymer (amylopectin). Amylose, usually the minor component of starch, is a long linear polymer containing 250 to 2000 D-glucose units connected by α -1,4-linkages, with a corresponding molecular weight of approximately 4000-340000 (Glicksman, 1969).

Amylopectin is a highly branched, treelike configuration composed of linear chains similar to those of amylose, but at branch points connected by α -1,6-linkages. These branch points are believed to occur at intervals of about 20-30 glucose units. The total amylopectin molecule is composed of several hundred branches and molecular weight of amylopectin is considered to be in the millions. Amylopectin has a globular shape that shows enhanced dilation and higher viscosities in the solution (Glicksman, 1969).

A number of modified starches can be used with a wide range of hydration and film-forming characteristics. Pre-gelatinization is the simplest modification. Until gelatinization is achieved the starch is heated in water and then dried to a power to obtain pre-gelatinized starch. Extensive modifications including changes in the degree of branching (variations in amylose and amylopectin content) can be accomplished (Loewe, 1990).

Gelatinization is the phenomenon shown by starches when they are heated in aqueous dispersion. When an aqueous suspension of starch is heated, a temperature is reached at which the hydrogen-bonding forces are weakened so that water can be absorbed by the granules. As the temperature of the aqueous suspension of starch is raised hydrogen bonds continue to be disrupted, water molecules become attached to the hydroxyl groups and the granules continue to swell. With continued swelling of the granules, starch molecules that have become fully hydrated separate from the intricate network and diffuse into the surrounding aqueous medium (Glicksman, 1969). Amylose is considered primarily responsible for gel formation. It is the chief material that forms gel network, which binds and entraps unabsorbed water. It also links together intact starch granules or fragments thus providing additional structure in the network (Ott and Hester, 1965).

Corn starch, for example, is different from wheat starch in terms of the size and shape of their granules, so that their gelatinization properties, water absorption rate and swelling capacity are not the same (Fiszman and Salvador, 2003). In addition, the amount of leached amylose from the granules, responsible for the network formation, is higher for corn starch (Rovedo et al., 1999).

In general terms, starches can be divided into three types: those from roots (e.g., tapioca starch), those from tubers (e.g., potato starch) and those from cereals (e.g., wheat starch, corn starch, rice starch) (Sanderson, 1981).

Starches contain both the amylose and amylopectin polymers, the relative proportions of which are constant in any particular species of starch. The amylose and amylopectin content of different starches were given in Table 1.3.

Starch	Source	Amylose (%)	Amylopectin (%)
Wheat	cereal	23-27%	73-77%
Corn	cereal	24-28%	72-76%
Tapioca	root	17-20%	80-83%

Table1.3 Amylose and amylopectin content of starches (Zallie, 1988).

Increasing the amylose content would increase the polysaccharidepolysaccharide interaction, which gives a more crunchy batter and reduced oil uptake. However, too much amylose causes a fried product that is too hard/though to chew. It was reported that crispness is positively correlated with amylose content, while oil absorption is negatively correlated with amylose content (Mohammed et al., 1998). Amylose is known to form coherent and relatively strong in contrast to amylopectin films, which are brittle and non-continuous (Gennadios et al., 1997). Therefore, the amylose/amylopectin ratio in the batter formulation is important to determine product quality.

Dextrin has the same general formula as starch but a smaller and less complex molecule than any one of a number of carbohydrates. They are polysaccharides and are produced as intermediate products in the hydrolysis of starch by heat, by acids, and by enzymes. Their nature and their chemical behavior depend to a great extent on the kind of starch from which they are derived. For commercial use dextrin is prepared by heating dry starch or starch treated with acids to produce a colorless or yellowish, tasteless, odorless powder (http://www.infoplease.com/ce6/sci/A0815381.html).

Dextrins generally have a medium-high viscosity and help to the formation of a continuous, uniform batter (Fiszman and Salvador, 2003). The use of dextrins in batter formulations is related to an improvement in the crispness of the fried product (Shinsato et al., 1999).

1.4.2 Proteins

Formation of films from plant proteins such as wheat gluten and corn zein is important in affecting product quality. Film formation from corn zein, the prolamin fraction of corn proteins, and from wheat gluten, a mixture of the prolamin and glutenin fractions of wheat proteins play important role (Gennadios et al., 1997).

The amount and type of protein in flour affects the final product. In wheat flour, the proteins responsible for developing batter's characteristic structure are gliadin and glutenin. When water is added to the flour, these hydrate to form gluten. This is a strong elastic substance, which forms a network throughout the dough. The network traps carbon dioxide, produced by the added yeast and allows the dough to rise. The process of kneading dough helps develop the gluten network (http://www.nutrition.org.uk/information/foodandingredients/cereal.html).

Hard wheat flours, because of their higher protein content, require more water than soft wheat flours to yield comparable viscosities when used in a batter. This is due to the efficient water-binding capacity of the gluten protein (Loewe, 1990). Gnanasambandam and Zayas (1992) reported that batters containing wheat germ flour and corn germ protein flour improved batter characteristics by increasing water binding capacity and decreasing cooking loss.

1.4.3 Gums

Many of hydrocolloid substances used as ingredients in batters known as gums. They control viscosity and water holding capacity of batters. Some gums have the ability to participate in a gel or film formation in conjunction with other ingredients (Loewe, 1990).

1.4.3.1 Cellulose Derivatives

Cellulose chemically differs from starch simply by having β -1,4-linked rather than α -1,4-linked glucose units (Sanderson, 1981).

One of the most widely used of all gums, cellulose gums are a family of products made by chemically modifying cellulose. By this way cellulose, the long chain polymer found in most land plants, becomes water-soluble. Compounds such as carboxymethylcellulose (CMC), methylcellulose (MC) and hydroxypro-pylmethylcellulose (HPMC) are examples of modified celluloses (Dziezak, 1991). Treating cellulose with alkali to swell the structure, followed by reaction with propylene oxide and methyl chloride yields HPMC (Kester and Fennema, 1986).

MC and HPMC are the only gums that gel when heated and return to their original viscosities when they are cooled. This unusual property makes these gums suitable for use in fried foods in where they create a barrier to oil absorption by the product. They retard the loss of natural product moisture and improve the adhesion of batter to the product (Dziezak, 1991). In addition, these derivatives can function as emulsifiers and their acid stability is good since they do not contain negatively charged groups (Sanderson, 1981). The number of substituent groups on the ring determines properties of the product.

The use of HPMC in fried foods has been studied by a number of authors. Chicken balls coated with an HPMC edible film showed a reduction in oil absorption in the surface layer and the core, as well as an increase in moisture retention (Balasubramaniam et al., 1997). Meyers and Conklin (1990) reported that the effectiveness of HPMC to reduce oil absorption in fried battered products such as chicken pieces, fish and vegetables.

1.4.3.2 Xanthan Gum

Xanthan gum is the only microbial polysaccharide permitted in food. Culturing on a carbohydrate medium the bacterium *Xanthomonas campestris* produces this high molecular weight polysaccharide. The gum has a cellulosic backbone with trisaccharide branches attached to every second glucose unit. Xanthan gum is completely soluble in cold or hot water by the presence of these short side chains (Sanderson, 1981).

Xanthan gum has found use in many products for its thickening, suspending and stabilizing properties (Dziezak, 1991). Its water binding capacity is important for the batter systems to yield enhanced moisture retention within the product. Xanthan gum is an example of gum that is nonionic and is not affected by the presence of salt in the coating material (Loewe, 1990). Altunakar (2003) reported 1% xanthan gum addition to the tempura type batter provided significant decrease in oil uptake of chicken nuggets while affecting volume development within the fried product.

1.4.3.3 Guar Gum

Guar gum is obtained from the ground endosperm of the guar plant, *Cyamopsis tetragonolobus*. The backbone of the gum is a linear chain of mannose units. One galactose unit is attached as side chains per every two-mannose units of guar gum. Mannan backbone can be solubilized by the presence of single unit galactose side chains. The fact that guar gum can be dissolved in cold water while another galactomannan locust beam gum requires hot water is due to the more substituent structure of guar gum (Sanderson, 1981).

Guar gum is non-gelling but gives highly viscous solutions at low concentrations. Therefore, it is chiefly used as viscosity builder, stabilizer and water binder. In addition, since it is nonionic, it is not adversely affected by the presence of salt in the batter formulation (Dziezak, 1991). Patil et al. (2001) used guar gum (0.25-1%) in batter formulations and showed that 9.7 - 22% oil content reduction over the control during frying.

1.4.3.4 Synergistic Interactions

Starches and gums are often used together in food systems to provide proper texture, to control moisture, to improve overall product quality and to reduce costs (Shi and BeMiller, 2002).

Xanthan gum interacts synergistically with galactomannas, such as guar gum. The content of galactose and the distribution of galactose residues in the galactomannan can have a significant influence on the interaction with xanthan gum molecules (Tako, 1991). Mixtures of guar gum and xanthan gum do not normally gel but show significantly higher viscosities compared to the viscosities of sole components (Katzbauer, 1998).

Synergistic interactions also take place between starch and xanthan gum or guar gum. It is shown to be advantageous in order to obtain increased moisture retention (Katzbauer, 1998). Carlson et al. (1962) have reported that the viscosity of a combination of guar and wheat starch cooked at high temperature is higher than the total thickening capacity of individual ingredients. It is supposed that the wheat starch is tied to the guar gum by means of hydrogen bonding (Glicksman, 1969).

1.5 Objectives of the Study

Battered foods such as fish and poultry are very popular in the market. Very little technical literature exists on the application of batters and breadings to vegetable products. Onions are the most commonly coated vegetable. Other battered or breaded vegetable products are bell peppers, cauliflower, eggplant, mushrooms, okra, and zucchini. Nevertheless, research on carrots has not yet been reported.

The objectives of the study were to evaluate the effect of starch or gum added batter coatings on product quality during deep-fat frying of carrot slices. In the first part of the study, dextrin and pre-gelatinized tapioca starch at different concentrations (1%, 3% and 5%) were included in the batter formulation and effects of these ingredients on quality of fried carrot slices were investigated. Subsequently, the effects of HPMC, guar gum, xanthan gum and a combination of guar and xanthan gum in deep-fat frying of battered carrots were studied. In addition, contributions of gums and starches to the coating formulation in terms of fried product quality were compared and optimum frying time was reported. It was also aimed to analyze oil content of samples using Image processing.

CHAPTER 2

MATERIALS AND METHODS

2.1.Materials

Fresh good quality carrots (*Daucus carota* L.) were procured from a local super market. They were stored at 4°C prior the experimental runs. Initial moisture content of carrots was determined using AOAC Method 14.003 (AOAC 1980).

The hydrocolloids used in the study and their sources were dextrin (acid hydrolyzed, Başar Trade Company, Turkey), pre-gelatinized tapioca starch (Ultra-Tex TM 3, National Starch and Chemical Company, USA), hydroxypropylmethylcellulose (HPMC) (Methocel K, The Dow Chemical Company, USA), xanthan gum (Aldrich Chemical Company, USA) and guar gum (Aldrich Chemical Company, USA). Detailed information about HPMC can be obtained from <u>www.dow.com/methocel/resource/chem.htm</u> for Methocel K.

The other ingredients used in batter preparation and their sources were wheat flour (Pınar Un, Turkey), corn flour (Bünsa Trade Company, Turkey), salt (Billur Tuz, Turkey) and leavening agent (Kenton, Turkey).

The carrot slices were deep-fried in refined sunflower oil (Bizim Ayçiçek, Turkey), in an electric fryer (HAD, Turkey).
2.2. Batter Preparation

Dry solid content of control batter formulation was composed of equal amount of wheat and corn flour (49.25% each), 1% salt and 0.5% leavening agent.

The effects of dextrin and pre-gelatinized tapioca starch at different concentrations (1%, 3% or 5%) were studied by replacing wheat and corn flour mixture.

In the case of gums, 1% of wheat and corn flour mix was replaced with gums to study their effects on product quality. The gums used were hydroxypropylmethylcellulose (HPMC), xanthan gum, guar gum or the combination of guar and xanthan gums (0.5% each).

The pre-blended powders were mixed with water (15 °C) in a mixer (Arçelik ARK55 MS, Turkey) at speed 1 for 15 seconds for batter preparation. The proportion of dry mix / water was always 3 / 4.5.

2.3.Sample Preparation and Frying Conditions

Carrots were peeled and cut into slices of 60 mm x 30 mm x 2.7 mm by means of a manual peeler and slicer. They were immediately dipped in the coating suspensions for 5 s and then fried in a controlled temperature deep-fat fryer filled with 2.5 L of sunflower oil. Frying temperature was set at 170 ± 2 °C and temperature was monitored by a copper constantan thermocouple. This temperature was decided to be suitable according to the preliminary experiments. Batches of four battered slices were fried for 2, 3 or 4 minutes. The fried carrots were removed from the fryer, drained and allowed to cool to room temperature. After each frying, the oil level was checked and replenished; the oil was changed after 6 h of frying time.

2.4. Analysis of Sample

2.4.1.Coating Pick-Up Calculations

Batter pick-up was calculated from the difference between battered weight and non-coated weight of raw carrot sample. It can be formulated as in equation 2.1 (Parinyasiri et al., 1991);

% Coating Pick-Up =
$$(C-I)/I * 100$$
 (2.1)

Where; C: weight of raw coated carrot slices (g)

I: initial weight of raw non-coated carrot slices (g)

2.4.2.Moisture Analysis

Moisture content was determined by measuring weight loss of fried products, upon drying in an oven at 105 °C until constant weight (AOAC, 1980). It was expressed as percentage of original sample.

2.4.3.Oil Analysis

Soxhlet extraction of the sample previously dried for moisture analysis was utilized to measure oil content. Extraction was performed with n-hexane for 6 hours and the oil content of sample was expressed as percentage of original sample (AOAC, 1984).

2.4.4.Frying Yield Calculations

Percentage of frying yield was obtained by considering the weights of the fried carrot slices and the raw carrot slices after coating. It can be formulated as equation 2.2 (Parinyasiri et al., 1991);

% Frying Yield =
$$(CW / C) * 100$$
 (2.2)

Where; CW: cooked weight of coated carrot slices (g)

C: weight of non-cooked coated carrot slices (g)

2.4.5.Colorimetric Measurements

The color parameters (Hunter L, a, b) were measured with a Minolta color reader (CR-10, Japan). The three color coordinates ranged from L=0 (black) to L=100 (white), -a (greenness) to +a (redness), and -b (blueness) to +b (yellowness) (Clydesdale 1984). Total color difference (ΔE) was calculated from Equation 2.3 (Ling et al., 1998);

$$\Delta E = [(L-L_{\text{standard}})^2 + (a-a_{\text{standard}})^2 + (b-b_{\text{standard}})^2]^{1/2}$$
(2.3)

Where; standard values referred to the BaCl₂ plate (L=96.9, a=0 and b=7.2) used for calibrating the colorimeter.

Triplicate readings were carried out at room temperature on three different locations of each slice: the center point and both ends, and the mean values were recorded.

2.4.6. True and Bulk Volume Measurements

True volume was measured by a stereopycnometer (Quantacrome, USA) using nitrogen. A tank pressure of 1.406 kgf/cm² was used. True volume was calculated from equation 2.4.

$$V_{t} = V_{2} + V_{1} \left[\left(P_{2} - P_{1} \right) / P_{2} \right]$$
(2.4)

Where; V_t : true volume of carrot slices (cm³)

 V_1 : volume of the first chamber (cm³)

 V_2 : volume of the second chamber (cm³)

 P_1 : equilibrium pressure when the second chamber is closed (kgf/ cm²)

 P_2 : equilibrium pressure when the second chamber is opened (kgf/ cm²)

Bulk volume of sample was measured by immersing the samples in a graduated cylinder filled with paraffin. Bulk volume was calculated from equation 2.5.

$$V_{b} = [(W_{pf} - W_{p}) - (W_{pfs} - W_{ps})] / \rho_{f}$$
(2.5)

Where; V_b: bulk volume of carrot slices (cm³)

W_{pf}: weight of the pycnometer filled with paraffin (g)

W_p: weight of empty pycnometer (g)

 W_{pfs} : weight of the pycnometer containing the sample and filled with paraffin (g).

 W_{ps} : weight of the pycnometer containing sample with no paraffin (g) ρ_{f} : density of the paraffin at 25 °C (g/ cm³)

2.4.7. True and Bulk Density Calculations

True and bulk densities of the sample were determined by dividing weight of the sample by true and bulk volume, respectively.

The main reason to calculate density may be expressed by the fact that density serves as the first estimate to porosity (Marousis and Saravacos, 1990).

2.4.8.Porosity Calculations

Porosity (ϵ) defined as the volume fraction of air or void fraction in the sample and was determined from equation 2.6 (Pinthus et al., 1995a);

$$\varepsilon = 1 - (\rho_b / \rho_t) \qquad (2.6)$$

Where; ρ_b : bulk density (g/cm³)

 ρ_{t} : true density (g/cm³)

2.4.9. Textural Measurements

A standard texture analyzer (Lloyd Instruments, TA Plus, U.K.) was used to evaluate the fracturability of products. Penetrometry tests were performed at 15 minutes after frying. Three carrot slices were put on top of each other and the thickness of fried carrot slices were detected with a micrometer (Mitutoyo, Japan). A conic plunger (D=1.6 cm, H=1.5 cm) was utilized to measure the force required to penetrate 25% thickness of products. A load cell of 50 N was used.

2.4.10.Image Processing

The fried samples were cut into rectangular shapes 1.20 x 2.30 cm in size and the batter separated from the carrot portion with a scalpel. Special care was taken not to remove oil droplets, which were also visible with the naked eye. After separation, the carrot and batter sections were placed on to the slides and observations were carried out with a digital camera (Kodak DX4530 5 megapixel CCD sensor, 5.2 megapixel CCD resolution and 5 megapixel image resolution). Images were enlarged 10 times of their original size and image analysis was performed using Image J software.

2.4.11. Statistical Analysis

All measurements were performed at least in triplicate and mean values were reported. Analysis of variance (ANOVA) was performed to study differences in quality parameters of deep-fat fried carrot slices coated with different formulated tempura batters. When significant differences were found the Duncan's Multiple Comparison test was applied to determine the difference among means ($p \le 0.05$) (SAS, 1988).

Correlations were obtained to relate moisture content to oil content and moisture content to frying yield.

CHAPTER 3

RESULTS AND DISCUSSION

3.1.Effects of Tapioca Starch and Dextrin on the Quality Parameters of Deep-fat Fried Carrot Slices

Initial moisture contents of the carrots were determined to be 90 ± 1 %. The effects of adding pre-gelatinized tapioca starch or dextrin to the batter formulation at different concentrations (1, 3, 5%) on coating pick-up and on major quality parameters of fried carrot slices; moisture and oil contents, frying yield, crispness, porosity and color were investigated. No starch or dextrin added batter was used as control batter formulation.

3.1.1.Coating Pick-up

Coating pick-up is an important physical property since it affects quality parameters of fried products. Coating pick-up for the batters containing different concentrations of pre-gelatinized tapioca starch or dextrin is shown in Figure 3.1.

The increasing concentrations of pre-gelatinized tapioca starch gave higher coating pick-up values (Figure 3.1). Concentration of 3% and 5% tapioca starch gave significantly higher batter pick-up values as compared to control and dextrin containing batters (Table B.1). This may be explained by high water binding capacity of pre-gelatinized tapioca starch (Appendix C). Batter viscosity is the key factor to control the amount of batter pick-up. A good correlation between batter viscosity and pick-up was previously reported (Altunakar, 2003; Dogan, 2004). Pre-gelatinized starch increased batter pick-up by its viscosity enhancing property.

The addition of dextrin at different concentrations (1, 3, 5%) didn't affect coating pick-up in the studied concentration range (Table B.1). Similar result was observed for dextrin at the concentrations of 1.5, 4.5 and 7.5 % by Baixauli et al. (2003).



Figure 3.1 Effects of pre-gelatinized tapioca starch and dextrin at different concentrations on coating pick-up of fried carrot slices.

(1) control, (2) dextrin 1%, (3) dextrin 3%, (4) dextrin 5%, (5) tapioca starch 1%, (6) tapioca starch 3%. (7) tapioca starch 5%.

* means bars with different letters are significantly different ($p \le 0.05$).

3.1.2. Moisture Content

The moisture content of the fried carrot slices was affected by different formulations. As expected, moisture content of products decreased with frying time (Figure 3.2).

The role of pre-gelatinized tapioca starch addition is to complement or improve the properties of the natural starches in the flour fraction of batter (Davis, 1983). Data indicated that samples coated with pre-gelatinized tapioca starch added batter enabled high moisture retention within the product (Figure 3.2). It is related with high coating pick-up values of this type of batter (Figure 3.1).

In Figure 3.2 it is seen that increasing dextrin concentrations has an adverse effect on moisture retention. Addition of 3% or less dextrin to the batter formulation improved moisture retention (Table B.2). At higher concentrations, especially at 3 and 4 minutes of frying, dextrin addition to the batter mix had no advantage in terms of controlling moisture loss. This may be due to the dilution of gluten in batter formulation.



Figure 3.2 Effects of pre-gelatinized tapioca starch and dextrin at different concentrations on moisture content of deep-fat fried carrot slices. (\Box) control^d, (-) dextrin 1%^b, (\blacklozenge) dextrin 3%^c, (*)dextrin 5%^{cd}, (o) tapioca starch 1%^b, (\diamondsuit) tapioca starch 3 %^{ab}, (x) tapioca starch 5%^a.

3.1.3. Oil Content

Oil contents of carrot slices during frying were represented in Figure 3.3. Oil absorption of products increased as the retention time of slices in frying medium increased. An inverse relationship was seen between the oil uptake and moisture content of fried carrots (Figure 3.2 and Figure 3.3). Since the samples had the same initial water content, an increase in final moisture content means a reduction in moisture loss during frying, which is normally correlated with oil uptake for the fried product. A correlation was determined between oil content and moisture content (r= -0.88). It was known that batter coating functions to reduce water loss which, in turn, lessens oil absorption during frying (Mohammed et al., 1998).

The water binding capacity of pre-gelatinized tapioca starch affected simultaneously water loss and oil uptake. The increase in concentrations of pre-gelatinized tapioca starch decreased oil uptake values (Figure 3.3). Usage of starch was suitable for reducing oil uptake since it was hydrophilic and in readily gelatinized form. The lowest oil uptake was determined at 5% pre-gelatinized tapioca starch concentration, which was related with its high moisture retention capability (Table B.3). A high concentration of starch is necessary to yield desirable product quality in means of high moisture and low oil uptake.



Figure 3.3 Effects of pre-gelatinized tapioca starch and dextrin at different concentrations on oil uptake of deep-fat fried carrot slices. (\Box) control^a, (-) dextrin 1%^b, (\blacklozenge) dextrin 3%^a, (*)dextrin 5%^a,

(o) tapioca starch $1\%^{b}$, (\Diamond) tapioca starch $3\%^{b}$, (x) tapioca starch $5\%^{c}$.

Addition of 1% dextrin enabled resistance to transport of water vapor and so decreased oil uptake. On the other hand, higher concentrations of dextrin addition levels dilute protein content of coating material. Therefore, the moisture retention capacity of batter decreased and more oil absorption by the product was observed. The dilution problem of proteins might not be compensated by addition of dextrin, a low molecular weight polysaccharide.

3.1.4. Frying Yield

Percentage frying yield is related with the weight change of coated samples during frying (Parinyarisi, 1991). It also indicates adhesion during frying which is important in terms of economic feasibility. It might be evaluated with simultaneous moisture loss and oil uptake mechanism of deep fat frying, which causes respectively weight loss and gain during frying. The decrease in frying yield with time show that the rate of moisture loss is higher than that of oil uptake. The observed high frying yield (Figure 3.4 and Table B.4) with addition of 3% and 5 % pre-gelatinized tapioca starch could be due to increased amount of moisture retention within the product. Frying yield was found to be related with moisture content and correlation coefficient was determined to be 0.90.



Figure 3.4 Effects of pre-gelatinized tapioca starch and dextrin at different concentrations on frying yield of deep-fat fried carrot slices. (\Box) control^c, (-) dextrin 1%^{bc}, (\blacklozenge) dextrin 3%^b, (*)dextrin 5%^{bc}, (o) tapioca starch 1%^b, (\diamond) tapioca starch 3 %^a, (x) tapioca starch 5%^a.

3.1.5.Crispness

One of the most appreciated characteristics in battered and fried products is an external crispy crust without being very hard (Salvador et al., 2002). Therefore, the textural performance of the product was associated with the fracturability data analysis, which is a good indicator of crispness. The typical texture profile analysis (TPA) curve can be seen in Appendix A.

Adding 5% pre-gelatinized tapioca starch to the formulation increased crispness of samples significantly (Figure 3.5 and Table B.5). Texture of the fried batters is influenced by the degree of polysaccharide-polysaccharide, polysaccharide-protein, polysaccharide-water and polysaccharide-oil interaction. The ability of the branched amylopectin structure to hold and interact strongly

with water resulted in a soft soggy batter (Mohammed et al., 1998). The amylose content of starches enhanced the polysaccharide-polysaccharide interactions and gave crispness to the crust of products. The usage of increasing concentrations of pre-gelatinized tapioca starch provided higher fracturability values (Figure 3.5). These values also enhanced with frying time. During frying process the swelling of starch granules releases the amylose fraction and provides a film barrier. Gelatinization and the film formation play an important role in crispy structure of the finished product (Arenson, 1969).

Replacing part of the wheat and corn flour with dextrin also changed sample's fracturability (Figure 3.5 and Table B.5). An increase in fracturability was obtained when high concentrations of dextrin were added to the batter formulation. This is probably because of the reduction in protein content and so the water binding capacity. Similar findings about the effect of dextrin on the crispness of the coated fried foods have been reported in the literature (Shinsato et al., 1999). The addition of dextrin to the coating batter of squid rings produced crisp texture and the texture was retained longer after frying (Baixauli et al., 2003).

Considering the later stages of frying, 5% pre-gelatinized tapioca starch or 5% dextrin addition to the batter formulation provided high crispness since the acceptable product can be obtained between 3 and 4 minutes.



Figure 3.5 Effects of pre-gelatinized tapioca starch and dextrin at different concentrations on fracturability of deep-fat fried carrot slices. (\Box) control^e, (-) dextrin 1%^{de}, (\blacklozenge) dextrin 3%^{cd}, (*)dextrin 5%^b, (o) tapioca starch 1%^c, (\diamond) tapioca starch 3 %^{bc}, (x) tapioca starch 5%^a.

3.1.6. Bulk Density

In Figure 3.6 the bulk densities of the fried products coated with batter containing pre-gelatinized tapioca starch or dextrin at different concentrations were given. The bulk densities of fried products with pre-gelatinized tapioca starch or dextrin added batters were markedly lower than the control batter coated products (Table B.6). As mentioned before with increasing pre-gelatinized tapioca starch concentration, batter pick-up increased, which enhanced the formation of a hard crust during frying (Figure 3.1 and 3.5). The crust serves as a barrier to prevent water loss and, as a result, contributes to reduction in oil absorption (Shih and Daigle, 1999). The crust is also responsible for gas retention within the product. This explains the lower bulk density or higher specific bulk volume of fried carrot slices coated with batter containing pre-gelatinized tapioca starch.

Dextrin addition to the batter formulation also reduced bulk density of the fried product significantly but its effect was lower as compared to that of pregelatinized tapioca starch added formulations (Table B. 6).

The decrease in the bulk density with respect to time was the result of expansion of carrot slices during deep-fat frying process. As can be seen in Figure 3.6, the rate of decrease in bulk density during frying was higher in carrot slices coated with control and dextrin added batter formulations. This was presumably the result of expansion caused by sudden loss of moisture from slices.



Figure 3.6 Effects of pre-gelatinized tapioca starch and dextrin at different concentrations on bulk density of deep-fat fried carrot slices. (\Box) control^a, (-) dextrin 1%^{ab}, (\blacklozenge) dextrin 3%^{bc}, (*)dextrin 5%^c, (o) tapioca starch 1%^d, (\diamond) tapioca starch 3 %^d, (x) tapioca starch 5%^d.

3.1.7. Porosity

Crust porosity enhanced linearly with frying time and positively affected by adding pre-gelatinized starch or dextrin to the coating material (Figure 3.7).

Duncan's multiple range test also showed that using pre-gelatinized tapioca starch or dextrin enabled more porous products than control batter (Table B.7). Addition of 5% pre-gelatinized tapioca starch provided the most porous structure. This observation can be evaluated by less oil uptake into pores of slices during frying process since the pores of the crust were not filled with frying oil.



Figure 3.7 Effects of pre-gelatinized tapioca starch and dextrin at different concentrations on porosity of deep-fat fried carrot slices. (\Box) control^e, (-) dextrin 1%^{bc}, (\blacklozenge) dextrin 3%^{cd}, (*)dextrin 5%^d, (o) tapioca starch 1%^{ab}, (\diamondsuit) tapioca starch 3 %^{ab}, (x) tapioca starch 5%^a.

3.1.8. Color

The final color of the fried product depends on the frying time, the absorption of oil and the chemical browning reactions of reducing sugars and protein sources (Baixauli et al., 2002). Lightness value decreased while Hunter a value increased during frying (Figure 3.8 and Figure 3.9). There was no definite trend in variation of Hunter b value during frying (Figure 3.10). Significantly higher Hunter L values were obtained when pre-gelatinized tapioca starch was added to the formulation. It is probably related with the reduced amount of oil uptake during frying process (Loewe, 1990).





(□) control^a, (-) dextrin 1%^{ab}, (◆) dextrin 3%^{bc}, (*)dextrin 5%^c,
(o) tapioca starch 1%^d, (◊) tapioca starch 3 %^d, (x) tapioca starch 5%^d.

Increased amount of dextrin concentration resulted in lower lightness values (darker color) within the product, which is due to the increase in the rate of Maillard and caramelization reactions (Table B.8).





(□) control^d, (-) dextrin 1%^c, (◆) dextrin 3%^b, (*)dextrin 5%^a,
(o) tapioca starch 1%^{de}, (◊) tapioca starch 3%^e, (x) tapioca starch 5%^e.

In Figure 3.9 Hunter a values of fried carrot slices were represented. Positive Hunter a values represented redness of products. No significant difference was observed between pre-gelatinized tapioca starch concentrations (Table B.9). However significant differences were reported between dextrin concentrations (Table B.9). Addition of 5% dextrin concentration provided the highest a value (highest redness) due to the contribution of dextrin in non-enzymatic browning reactions.





- (\Box) control, (-) dextrin 1%, (\blacklozenge) dextrin 3%, (*) dextrin 5%,
- (o) tapioca starch 1%, (\Diamond) tapioca starch 3 %, (x) tapioca starch 5%.

In Figure 3.10, Hunter b values of fried carrot slices were represented. Positive b value represented yellowness of products. As mentioned before, a definite trend could not obtained for b values of deep-fat fried carrot slices.



Figure 3.11 Effects of pre-gelatinized tapioca starch and dextrin at different concentrations on ΔE of deep-fat fried carrot slices. (\Box) control^{bc}, (-) dextrin 1%^{bc}, (\blacklozenge) dextrin 3%^b, (*)dextrin 5%^a, (o) tapioca starch 1%^{bc}, (\diamondsuit) tapioca starch 3%^{cd}, (x) tapioca starch 5%^d.

Color change (ΔE) of carrot slices was also given in Figure 3.11. Duncan's multiple range test showed that 5% dextrin addition to the formulation provided higher ΔE value during frying (Table B.10).

3.2.Effects of Different Gum Types on the Quality Parameters of Deep-fat Fried Carrot Slices

The effects of different gum types (HPMC, guar gum, xanthan gum and guar-xanthan gum combination) on coating pick-up and on major quality parameters of fried products; moisture and oil contents, crispness, porosity, color and frying yield were evaluated in this part of the study. To determine the effectiveness of gums no gum added coating was used as the control batter formulation.

3.2.1. Coating Pick-up

Batter pick-up is generally directly correlated with batter viscosity: that is, as viscosity increases, more batter remains on the sample (Cunningham and Tiede, 1981; Altunakar, 2003; Dogan, 2004). Gums are able to provide high viscosity to their dispersions even at low concentrations. Therefore, gum addition to the batter formulation resulted in higher batter pick-up values (Figure 3.12). No significant differences were detected between gum types but the combination of guar and xanthan gum yielded higher coating pick-up due to the high degree of synergism, which led them to suggest interactions between molecules as a possible cause of the increase in the viscosity (Table B.11). It is very well known that synergistic interaction of xanthan gum with guar gum gives a synergistic increase in viscosity (Sanderson, 1981).



Figure 3.12 Effects of gum types on coating pick-up of deep-fat fried carrot slices. (1) control, (2) HPMC, (3) xanthan gum, (4) guar gum, (5) guar-xanthan gum combination. * means bars with different letters are significantly different ($p \le 0.05$).

3.2.2. Moisture Content

Moisture contents of fried carrot slices coated with batters containing different gum types and with the control formulation were represented in Figure 3.13. Duncan's multiple range test showed that all types of gums were significantly effective in controlling moisture loss (Table B.12).

Film formation and thermal gelation abilities are critical functions of gums for both barrier properties; moisture retention and oil uptake reduction (Loewe, 1990). The higher moisture retention was reported when guar gum or the combination of guar and xanthan gum was used. This may be due to high water binding capacities of these gums (Table C.1). Synergistic interaction of guar gum with xanthan gum is important to increase viscosity (Dziezak, 1991). Mixtures of guar gum-xanthan gum do not usually help gelatinization but show enhanced batter pick-up values compared to values of individual components, which resulted in decreased moisture loss within the product (Figure 3.12 and Figure 3.13). A homogenous film formation provides efficient and effective coverage of the product in order to decrease moisture loss (Loewe, 1990).

Thermal gelation property of HPMC above its incipient gelation temperature is important in controlling moisture loss. The methyl groups in HPMC molecules undergo intermolecular association with adjacent molecules above gelation temperature. As a result, viscosity increases dramatically with increase in temperature (Mallikarjunan et al., 1997).



Figure 3.13 Effects of gum types on moisture content of deep-fat fried carrot slices. (\Box) control^c, (\blacktriangle) HPMC^b, (Δ) xanthan gum^b, (•)guar gum^a, (+) guar-xanthan gum combination^a.

3.2.3. Oil Content

The gelling ability of gums together with their usual hydrophilic nature and film forming ability makes them useful for reducing oil uptake during frying in battered products (Annapure et al., 1999). Oil contents of products provided with coatings containing different gum types and with the control formulation were shown in Figure 3.14.

The trend of oil uptake of carrot slices during frying was the reverse of that shown by moisture content. Using gums in batter formulations resulted in lower oil uptake by enhancing moisture retention as a result of a strong interaction due to hydrogen bonding between water molecules in the batter and gums. The displacement of water by oil during frying is restricted.



Figure 3.14 Effects of gum types on oil content of deep-fat fried carrot slices. (\Box) control^a, (\blacktriangle) HPMC^c, (Δ) xanthan gum^b, (\bullet)guar gum^d, (+) guar-xanthan gum combination^e.

Guar-xanthan gum added coating reduced oil uptake significantly as compared to other coatings (Table B.13). This can be explained by the synergistic effect, which led to more batter pick-up, resulting in high moisture retention and low oil content.

Thermo-gelling property of HPMC promotes the formation of a small amount of wide punctures with low capillary pressures (Mellema, 2003). Less oil uptake was observed in HPMC added batters as compared to the products coated with control batter. It is probably due to the low capillary pressure, which resulted in less oil entrance to the pores of slices. In literature, it was also reported that HPMC reinforces the natural barrier properties of starch and proteins especially when they are added in dry form (Myers, 1990).

3.2.4.Frying Yield

Percent frying yield is an indicator of batter adhesion during frying process (Hsia et al., 1992). All gum containing batters supplied yields greater than the control (Figure 3.15 and Table B.14). The higher frying yield values obtained for guar and xanthan gum combination added batter is probably because of high moisture retention. Besides batter adhesion, in calculating percent frying yield, both moisture loss and oil absorption plays role.



Figure 3.15 Effects of gum types on frying yield of deep-fat fried carrot slices. (\Box) control^d, (\blacktriangle) HPMC^{bc}, (Δ) xanthan gum^{ab}, (\bullet)guar gum^c, (+) guar-xanthan gum combination^a.

3.2.5.Crispness

Fracturability values of all gums were found to be significantly different from control batter (Figure 3.16 and Table B.15). It may be explained by thermogelling and cross-linking properties of gums.



Figure 3.16 Effects of gum types on fracturability of deep-fat fried carrot slices. (\Box) control^c, (\blacktriangle) HPMC^{ab}, (Δ) xanthan gum^{ab}, (\bullet)guar gum^a, (+) guar-xanthan gum combination^b.

3.2.6. Bulk Density

The variation of bulk densities of carrot slices coated with different batter formulations during frying were shown in Figure 3.17. The densities of carrot slices decreased during frying. Bulk densities of carrot slices were significantly reduced when gums were used in batter formulations (Table B.16). This is mainly due to better film forming and gas holding ability of gum added batters

Usage of xanthan and guar gum combination had no advantage over usage of xanthan gum only with respect to bulk density (Figure 3.17 and Table B. 16). However the combination may be preferred since xanthan gum is higher in price. The sharp decrease in bulk density within the initial period of frying in the case of control formulation was most probably due to the expansion caused by sudden moisture loss from the slices.



Figure 3.17 Effects of gum types on bulk density of deep-fat fried carrot slices. (\Box) control^a, (\blacktriangle) HPMC^b, (Δ) xanthan gum^c, (\bullet)guar gum^b, (+) guar-xanthan gum combination^c.

3.2.7. Porosity

The porosity data of experiments were shown in Figure 3.18. It is obviously seen that adding gums to the coating enables more porous products (Table B.17).

The differences between gum types can be explained with the different film forming and gas retention abilities of gums. During frying, oil can be taken up before the food is taken from the frying medium as in the case of small food pieces like thin carrot slices (Mellema, 2003). Therefore, the barrier property to oil uptake may help to prevent filling the voids of the crust enabling more porous product.

When xanthan gum was used in batter formulation highly porous product was obtained as compared to other formulations (Figure 3.14 and Figure 3.18). This may be due to the better film forming and so gas retaining ability of xanthan gum. Lower bulk density of samples coated with xanthan gum added batter formulation also confirms this fact (Figure 3.17).



Figure 3.18 Effects of gum types on porosity of deep-fat fried carrot slices. (\Box) control^d, (\blacktriangle) HPMC^c, (Δ) xanthan gum^a, (\bullet)guar gum^c, (+) guar-xanthan gum combination^b.

3.2.8. Color

Gum addition resulted in higher L but a lower a value meaning lighter and less red color (Figure 3.19, Figure 3.20, Table B.18 and Table B.19). In literature, it is reported that gum's ability to bind moisture prevents dehydration and inhibits the Maillard browning reaction. The lighter color can also indicate that the product absorbed less frying oil (Loewe, 1990). Therefore, it was not surprising that gum types provided significantly lighter colors to products.



Figure 3.19 Effects of gum types on Hunter L value of deep-fat fried carrot slices. (\Box) control^d, (\blacktriangle) HPMC^c, (Δ) xanthan gum^{ab}, (\bullet)guar gum^{bc}, (+) guar-xanthan gum combination^a.



Figure 3.20 Effects of gum types on Hunter a value of deep-fat fried carrot slices. (\Box) control^a, (\blacktriangle) HPMC^b, (Δ) xanthan gum^b, (\bullet)guar gum^b, (+) guar-xanthan gum combination^b.

Gum types enabled significantly lower Hunter a values in comparison with control batter (Figure 3.20). No significant difference was seen between gum types in terms of redness given to the product during frying process (Table B.19).



Figure 3.21 Effects of gum types on Hunter b value of deep-fat fried carrot slices. (\Box) control, (\blacktriangle) HPMC, (Δ) xanthan gum, (\bullet)guar gum, (+) guar-xanthan gum combination.

A definite trend was not observed for variation of Hunter b values of coated carrot slices with different gum types during frying (Figure 3.21). Color changes (ΔE values) during frying of carrots coated with different gums were significantly less than the ones coated with control batter (Figure 3.22).



Figure 3.22 Effects of gum types on ΔE of deep-fat fried carrot slices. (\Box) control^a, (\blacktriangle) HPMC^c, (Δ) xanthan gum^c, (\bullet)guar gum^b, (+) guar-xanthan gum combination^c.

3.3. Comparison of the Effects of Pre-gelatinized Tapioca Starch, Dextrin and Gums on Deep-Fat Fried Carrot Slices

The effects of pre-gelatinized tapioca starch & dextrin at concentrations of

1,3, 5% and different gums (HPMC, guar gum, xanthan gum and guar-xanthan gum combination) on some quality parameters of fried carrot slices were compared in this section. Frying time of 3 minutes was chosen to make this evaluation, since at this time acceptable products were obtained.



Figure 3.23 Effects of different hydrocolloids on coating pick-up of deep-fat fried carrot slices.

(1) control, (2) 1% dextrin, (3) 3% dextrin, (4) 5% dextrin, (5) 1% pre-gelatinized tapioca starch, (6) 3% pre-gelatinized tapioca starch, (7) 5% pre-gelatinized tapioca starch, (8) 1% HPMC, (9) 1% xanthan gum, (10) 1% guar gum, (11) 1% guar-xanthan gum combination.

* means bars with different letters are significantly different ($p \le 0.05$).

The coating pick-up data for different batter formulations were given in Figure 3.23. Usage of dextrin at all concentrations and pre-gelatinized tapioca starch at 1% concentration do not have any significant improvement on coating pick-up of batter (Table B.21). Significantly higher pick-up values were obtained for pre-gelatinized tapioca starch at concentrations higher than 1% and for all gum types especially for guar-xanthan gum combination (Table B.21). It is known that coating pick-up is positively correlated with batter viscosity. Therefore, the differences among pick-up values were most probably due to the difference in their viscosities. Gums are commonly used for cold-batter viscosity adjustment (Davis, 1983). As well as gums, starches provide viscosity and so coating pick-up,
but gums are effective at much lower concentration levels. This lower concentration provides gums to be more cost effective than starches (Davis, 1983).



Figure 3.24 Effects of different hydrocolloids on moisture content of deep-fat fried carrot slices.

(1) control, (2) 1% dextrin, (3) 3% dextrin, (4) 5% dextrin, (5) 1% pre-gelatinized tapioca starch, (6) 3% pre-gelatinized tapioca starch, (7) 5% pre-gelatinized tapioca starch, (8) 1% HPMC, (9) 1% xanthan gum, (10) 1% guar gum, (11) 1% guar-xanthan gum combination.

Moisture and oil contents of carrot slices fried for 3 minutes were shown in Figure 3.24 and Figure 3.25, respectively. Usage of hydrocolloids strengthens the coating and provides more moisture retention enabling low oil uptake during deep-fat frying by means of their better film forming and water binding abilities. The most effective ingredient for reducing moisture loss and so oil uptake was guar and xanthan gum combination (Table B.22 and Table B.23).

All of the hydrocolloids used in this study can be recommended for lower oil uptake except dextrin at higher concentrations.



Figure 3.25 Effects of different hydrocolloids on oil content of deep-fat fried carrot slices.

(1) control, (2) 1% dextrin, (3) 3% dextrin, (4) 5% dextrin, (5) 1% pre-gelatinized tapioca starch, (6) 3% pre-gelatinized tapioca starch, (7) 5% pre-gelatinized tapioca starch, (8) 1% HPMC, (9) 1% xanthan gum, (10) 1% guar gum, (11) 1% guar-xanthan gum combination.

One of the most appreciated characteristics in fried products is crispness that can be associated with fracturability data. Crispness data of fried samples were given in Figure 3.26. Slight differences were obtained among different formulations (Table B. 24). Crispness increased with the use of gums since they provide structural integrity to the batter coating during frying. The high crispness value for guar gum may be due to the synergistic interaction of guar gum with wheat starch (Carlson et al., 1962).



Figure 3.26 Effects of different hydrocolloids on fracturability of deep-fat fried carrot slices.

Bulk densities of coated products were compared in Figure 3.27. It is obviously seen that dextrins and control batter provided higher density, which means lower volume (Table B.25). The gas may not be kept within the system due to lower pick-up values of these formulations. Addition of pre-gelatinized tapioca starch or gums to the batter formulation improved film-forming ability, which is important for gas retention during leavening. As a result more aerated and porous structure is obtained. The low level of gum addition may not be significant in dilution of gluten in batter formulation. Therefore, gums especially HPMC and guar and xanthan gums in combination were much more effective in obtaining higher bulk volume.



Figure 3.27 Effects of different hydrocolloids on bulk density of deep-fat fried carrot slices.

Lightness data of fried products were shown in Figure 3.28. Gums and pregelatinized tapioca starch at high concentrations have the ability to bind more water, which inhibits Maillard reaction (Figure 3.24). Significantly lighter color that provided by gums and high concentrations of pre-gelatinized tapioca starch can also be due to less oil absorption within the product (Table B.26, Figure 3.25 and Figure 3.26).



Figure 3.28 Effects of different hydrocolloids on Hunter L value of deep-fat fried carrot slices.

Hunter a values of deep-fat fried carrot slices were represented in Figure 3.29. Dextrin addition to batter formulation significantly changed the redness of fried products (Table 3.27).



Figure 3.29 Effects of different hydrocolloids on Hunter a value of deep-fat fried carrot slices.

3.4. Image Analysis

Images of fried products coated with different batter formulations were discussed in this part of the study. After frying, oil droplets seen on the carrot and crust surface were detected by photographing and analyzed by the help of Image processing. The images were given in Appendix D. In Appendix E surface plots of images for control formulation and guar-xanthan gum combination were given as sample graphs. Lighter areas (oil droplets) in the source image represent lower elevations (valleys) while darker areas in the source image represent higher elevations (peaks). There is abundant proof that oil hardly penetrates in the cooked core (Pinthus et al., 1995a). Bouchon, et al. (2001) using infrared microspectroscopy showed that the oil penetration depth in potato is very close to the evaporation front. Therefore, so as to determine oil uptake of slices, oil droplets at the surface of the carrot and batter portion of the sample were detected with enlarging the surfaces.

For 3 and 4 minute-frying times no correlation was noticed between total oil fraction given by image process and gravimetrically obtained oil content data of products represented in Figure 3.3 and Figure 3.14. This might be due to the oil absorption to the inner parts of thin pieces for longer frying times. Hence to detect oil droplets only at the surface didn't represent the actual oil contents for later stages of frying.

On the other hand, image analysis results of 2 minutes fried carrot slices were positively correlated with oil content data of samples (Figure 3.3, Figure 3.14 and Table 3.1). The difference in different formulations on affecting oil uptake was batter reflected by the area fraction of oil droplets on carrot surface as compared to those on batter surface (Table 3.1). As can be seen in Table 3.1 total area fraction of oil droplets was lower in case of guar-xanthan gum combination added batter formulation. This is confirmed by the lowest oil content carrots coated with batters containing of guar-xanthan gum combination (Figure3.14).

Table 3.1 Area fraction and average size of droplets observed on surfaces of carrot and batter portions of the fried sample.

	Area fraction and average	Area fraction and average	Total
Additive	size of oil droplets observed on carrot surface	size of oil droplets observed on batter surface	area
			fraction
Control	$2.4\%; 0.000063 \text{ cm}^2$	$0.5\%; 0.000037 \text{ cm}^2$	2.9%
5% Dextrin	2.1%; 0.000056 cm ²	0.5%; 0.000035 cm ²	2.6%
1% Dextrin	1.9%; 0.000069 cm ²	0.5%; 0.000026 cm ²	2.4%
1% Starch	1.6%; 0.000047 cm ²	0.3%; 0.000026 cm ²	1.9%
Xanthan gum	1.4%; 0.000060 cm ²	0.2%; 0.000026 cm ²	1.6%
HPMC	0.7%; 0.000037 cm ²	0.4%; 0.000036 cm ²	1.1%
5% Starch	0.7%; 0.000034 cm ²	0.3%; 0.000027 cm ²	1.0%
Guar-xanthan	0.7%; 0.000042 cm ²	0.2%; 0.000026 cm ²	0.9%
gum combination			

CHAPTER 4

CONCLUSION AND RECOMMENDATIONS

Addition of gums to batter formulations increased moisture retention, crispness, frying yield and porosity but decreased oil uptake.

Guar-xanthan combination has been found to be the most effective additive on the batter performance. It provided the highest moisture retention, and lowest oil uptake within the product. In addition, the highest volume and lightest color were obtained when this combination was used. When carrot slices were dipped into guar-xanthan gum added batter formulation, fracturability values were higher than control but lower than the values obtained with other gums. Acceptable porosity values were observed during the usage of this gum combination.

Experimental results indicated that using pre-gelatinized tapioca starch at higher concentrations enhanced efficiency of batter while increasing dextrin concentrations had an adverse affect on the product quality. Usage of starch at relatively high levels, provided to obtain desirable products. Frying time for coated carrot slices can be recommended as 3 minutes.

The oil fraction obtained using image analysis of carrot and batter surfaces was correlated with oil content of samples during the initial frying period since oil hardly penetrates into the inner parts. However, this method of analysis was not useful in later stages of frying. The correlation coefficient between moisture content and frying yield was obtained as 0.90. Also, oil content was found to be related with moisture content and correlation coefficient was determined to be -0.88.

It was the first research made on deep-fat frying of coated carrot slices. For further research, different coating formulations can be evaluated. As an example, modified high amylose starch, extra wheat gluten or corn zein can be added to coatings of carrots. Different combinations of gums and starches can also be studied to determine their synergistic interactions.

To decrease the initial moisture content of carrots pre-drying or osmotic pre-treatment or pre-dust application can be used to improve product quality in deep- fat frying. Carrots having different shapes and thicker slices can be used. In these conditions optimum frying times have to be determined.

Unfortunately, sensory evaluation and carotene analysis could not be included in this study. Further researches will be helpful to understand deep-fat frying mechanism of coated carrot slices.

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APPENDIX A

TEXTURE PROFILE ANALYSIS



Figure A.1 Typical TPA curve (guar-xanthan gum combination coated carrot slices; 3 minutes fried).

APPENDIX B

ANOVA and DUNCAN TABLES

Table B.1 ANOVA and Duncan's Multiple Range Test Table for coating pick-upof fried samples with different concentrations of dextrin and tapioca starch.

Levels	Values									
7	control, 1% dextrin, 3% dextrin, 5% dextrin,									
	1% starch, 3% starch, 5% starch									
Number of observations in data set = 21										
Sum of										
Squares	Mean Square F value $P_r > F$									
1769.33631	294.88938 143.62 0.0001									
28.74560	2.05325									
1798.08191										
Type III S	S Mean Square F Value $P_r > F$									
1769,3363	1 294,88938 143,62 0.0001									
letter are not si	gnificantly different.									
Mean N	Formulations									
75.650 3	5% tapioca starch									
65.640 3	3% tapioca starch									
51.940 3	5% dextrin									
51.840 3	3% dextrin									
51.220 3	1% dextrin									
50.920 3	1% tapioca starch									
50.290 3	control									
	Levels 7 ons in data set = Sum of Squares 1769.33631 28.74560 1798.08191 Type III SS 1769,3363 letter are not si Mean N 75.650 3 65.640 3 51.940 3 51.840 3 51.840 3 51.220 3 50.290 3									

Table B.2 ANOVA and Duncan's Multiple Range Test Table for moisture content of fried samples with different concentrations of dextrin and pre-gelatinized tapioca starch.

Class		Levels		Values					
Formulations		7		control, 1% dextrin, 3% dextrin, 5% dextrin,					
				1% starch, 3% starch, 5% starch					
Frying time (min) 3				2, 3, 4					
Number of obs	ervati	ons in data s	et =	= 21					
Source DF		Sum of							
		Squares			Mean Square	F Value	$P_r > F$		
Model	8	413.107	16		51.63839	36.24	0.0001		
Error	12	17 100/1			1 42503				
Total	20	120 20758							
	20	150.207	50						
Source	DF	Type III SS		1	Mean Square	F Value	$P_r > F$		
Batter Type	6	158.624	66		26.43744	18.55	0.0001		
Frying time	2	254.482	50		127.24125	89.29	0.0001		
Duncan Grou	ping	Mean	N	-	Formulation	15			
А		54.1550	3		5% tapioca star	rch			
AB		52.7033	3		3% tapioca star	rch			
В		51.6000	3		1% dextrin				
В		51.3633	3		1% tapioca star	rch			
С		48.8367 3			3% dextrin				
CD		47.3600	3		5% dextrin				
D		45.9833	3		control				

Table B.3 ANOVA and Duncan's Multiple Range Test Table for oil content of fried samples with different concentrations of dextrin and pre-gelatinized tapioca starch.

Class	Levels	Val	Values					
Formulations	7	cont	control, 1% dextrin, 3% dextrin, 5% dextrin,					
		1% s	1% starch, 3% starch, 5% starch					
Frying time (min)	3	2, 3,	2, 3, 4					
Number of observ	ations in data set	= 21						
	Sum of							
Source D	F Squares		Mean Square	F Value	$P_r > F$			
Model 8	107.68878	3	13.46109	36.12	0.0001			
Error 1	2 4.47271	-	0.37272					
Total 20	0 112 1614)						
Source D	F Type III S	SS	Mean Square	F Value	$P_r > F$			
Batter Type 6	48.87122		8.14520	21.85	0.0001			
Frying time 2	58.81755		29.40877	78.90	0.0001			
Duncan Groupin	g Mean	N	Formulation	18				
А	14.9400	3	control					
А	14.7767	3	5% dextrin					
А	14.0667	3	3% dextrin					
В	12.5900	3	1% tapioca star	rch				
B			1% dextrin					
D	12.2100	3	1% dextrin					
B	12.2100 11.9167	3 3	1% dextrin 3% tapioca star	rch				

Table B.4 ANOVA and Duncan's Multiple Range Test Table for frying yield of samples with different concentrations of dextrin and pre-gelatinized tapioca starch.

С	lass		Levels		Values					
F	ormulations		7		cont	control, 1% dextrin, 3% dextrin, 5% dextrin,				
					1% s	starch, 3% starch, 5% starch				
Frying time (min)			3		2, 3,	. 4				
N	lumber of obs	ervati	ons in data	set =	= 21					
			Sum o	f						
	Source DF Squares		25		Mean Square	F Value	$P_r > F$			
	Model	8	382.18	3128		47 77266	31.43	0.0001		
	Frror	12	18 238	263		1 51988	51.15	0.0001		
	Total	12 20	400.41	002		1.51700				
-	Total	20	400.41	.992						
	Source	DF	Type I	Type III SS		Mean Square	F Value	$P_r > F$		
	Batter Type	6	123.50	1300	-	20 58300	13.54	0.0001		
		0	125.50	720		100.220	15.54	0.0001		
-	Frying time	2	258.67	//29		129.338	85.10	0.0001		
D	uncan Grouj	oing	Mean	N	I	Formulation	18			
	А		81.373	3		5% tapioca star	rch			
	А		79.600	3		3% tapioca star	rch			
	В		77.227	3		3% dextrin				
	В		77.007 3			1% tapioca star	rch			
	BC		75.577 3			1% dextrin				
	BC		11.9167	3		5% dextrin				
	С		73.953	3		control				

Table B.5 ANOVA and Duncan's Multiple Range Test Table for fracturability of fried samples with different concentrations of dextrin and pre-gelatinized tapioca starch.

Class		Levels		Values					
Formulations		7		control, 1% dextrin,	3% dextrin,	5% dextrin,			
				1% starch, 3% starch	starch, 3% starch, 5% starch				
Frying time (m	in)	3		2, 3, 4					
Number of obs	ervati	ons in data s	set =	= 21					
	DE	Sum of							
Source	DF	Squares		Mean Square	F Value	$P_r > F$			
Model	8	0.00000	032	0.00000004	35.34	0.0001			
Error	12	0.00000	001	0.000000008					
Total	20	0.00000033							
Source	DF	Type III SS		Mean Square	F Value	$P_r > F$			
Batter	6	0.0000	020	0.00000003	20.64	0.0001			
Туре	0	0.00000	020		29.04	0.0001			
Frying time	2	0.00000	012	0.0000006	52.97	0.0001			
Duncan Grou	ping	Mean	N	Formulation	18				
А		0.00547	3	5% tapioca star	ch				
В		0.00535	3	5% dextrin					
BC		0.00530	3	3% tapioca star	ch				
С		0.00527	3	1% tapioca star	ch				
CD		0.00524	3	3% dextrin	3% dextrin				
DE		0.00520	3	1% dextrin					
Е		0.00514	3	control	control				

Table B.6 ANOVA and Duncan's Multiple Range Test Table for bulk density of

 fried samples with different concentrations of dextrin and pre-gelatinized tapioca

 starch.

С	lass		Levels		Valı	Values				
F	ormulations		7		cont	control, 1% dextrin, 3% dextrin, 5% dextrin,				
					1% s	1% starch, 3% starch, 5% starch				
F	rying time (m	in)	3		2, 3,	. 4				
Number of observati			ons in data s	et =	= 21					
	~		Sum of							
	Source	DF	Squares		Mean Square		F Value	$P_r > F$		
-	Model	8	0.09654			0.01206	35.52	0.0001		
	Error	12	0.00419	1		0.00034				
	Total	20	0.10074							
_										
	Source	DF	Type III	SS	3	Mean Square	F Value	$P_r > F$		
-	Batter Type	6	0.07174			0.01195	34.20	0.0001		
	Frying time	2	0.02480	1		0.0124	35.47	0.0001		
D	uncan Grouj	ping	Mean	N	I	Formulation	ns			
	А		1.10333	3	}	control				
	AB		1.07700	3	;	1% dextrin				
	BC		1.05933	3	•	3% dextrin				
	С		1.04033	3	•	5% dextrin				
	D		0.97267	3	}	1% tapioca starch				
	D		0.96333	3	•	3% tapioca starch				

Table B.7 ANOVA and Duncan's Multiple Range Test Table for porosity of fried samples with different concentrations of dextrin and pre-gelatinized tapioca starch.

Class		Levels	Va	alues							
Formulations		7	co	control, 1% dextrin, 3% dextrin, 5% dextrin,							
			1%	1% starch, 3% starch, 5% starch							
Frying time (min	n)	3	2,	3, 4							
Number of obse	ervatio	ons in data se	t = 21								
		Sum of									
Source	DF	Squares		Mean Square	F Value	$\mathbf{h}^{\mathrm{L}} > \mathbf{h}$					
Model	8	0.02832		0.00354	20.81	0.0001					
Error	12	0.00204		0,00017							
Total	20	0.03036									
Source	DF	Type III	SS	Mean Square	F Value	$P_r > F$					
Batter Type	6	0.02317		0.00386	22.70	0.0001					
Frying time	2	0.00515		0.00257	15.14	0.0005					
Duncan Group	ing	Mean	N	Formulation	ns						
А		0.14675	3	5% tapioca star	rch						
AB		0.13560	3	3% tapioca star	rch						
AB		0.12661	3	1% tapioca sta	rch						
BC		0.11150	3	1% dextrin							
CD		0.08935	3	3% dextrin							
D		0.07323	3	5% dextrin							
E		0.04769	3	control							

Table B.8 ANOVA and Duncan's Multiple Range Test Table for Hunter L value of fried samples with different concentrations of dextrin and pre-gelatinized tapioca starch.

Class		Levels		Valı	Values					
Formulations		7		control, 1% dextrin, 3% dextrin, 5% dextrin,						
				1% s	tarch, 3% starch	, 5% starch				
Frying time (mi	3		2, 3,	4						
Number of obse	ervati	ons in data s								
		Sum of								
Source	DF	Squares		Mean Square		F Value	$P_r > F$			
Model	8	203.342	78		25.41784	37.85	0.0001			
Error	12	8.05854			0.671545					
Total	20	211.40132								
Source	DF	Type III	SS		Mean Square	F Value	$P_r > F$			
Batter Type	6	145.737	05		24.28950	36.17	0.0001			
Frying time	2	57.6057	2		28.80286	42.89	0.0001			
Duncan Group	ing	Mean	N		Formulation	15				
А		57.0067	3		5%tapioca star	ch				
В		55.3067	3		3% tapioca star	rch				
С		52.7033	3		control					
С		52.6167 3			1% tapioca star	rch				
С		51.4333 3			1% dextrin					
D		49.7733	3		3% dextrin					
D		49.0667	3		5 % dextrin					

Table B.9 ANOVA and Duncan's Multiple Range Test Table for Hunter a value of fried samples with different concentrations of dextrin and pre-gelatinized tapioca starch.

Class		Levels		Valu	ues				
Formulations		7		cont	control, 1% dextrin, 3% dextrin, 5% dextrin,				
				1% starch, 3% starch, 5% starch					
Frying time (mi	3		2, 3,	4					
Number of obse	ervati	ons in data s	et =	= 21					
Source DF		Sum of							
		Squares			Mean Square	F Value	$P_r > F$		
Model	8	226 343	73		28 29296	65 29	0.0001		
Error	12	5 20029	5 20020		0 43335		010001		
Total	20	231 544	02		0.15555				
	20	231.344	251.54402						
Source	DF	Type III	Type III SS		Mean Square	F Value	$P_r > F$		
Batter Type	6	188.690	36		31.44839	72.57	0.0001		
Frying time	2	37.6533	7		18.82668	43.44	0.0001		
Duncan Group	oing	Mean	N	-	Formulation	15			
A		15.4267	3		5 % dextrin				
В		13.8467	3		3% dextrin				
С		12.5100	3		1% dextrin				
D		9.2200	3		control				
DE		8.3233 3			1% tapioca star	rch			
Е		7.8833	3		3% tapioca starch				
E		7.2900	3		5%tapioca starch				

Table B.10 ANOVA and Duncan's Multiple Range Test Table for ΔE of fried samples with different concentrations of dextrin and pre-gelatinized tapioca starch.

	Levels		Values				
	7		control, 1% dextrin, 3% dextrin, 5% dextrin,				
			1% s	tarch, 3% starch	, 5% starch		
l)	3		2, 3,	4			
vatio	ons in data se	et =	= 21				
	Sum of						
DF	Squares			Mean Square	F Value	$P_r > F$	
8	166 3094	51		20 78868	24.00	0.0001	
17	10 20242))		0.86611	24.00	0.0001	
12 20	17(702)	2 22		0.00011			
20	1/6./0293						
DF	Turna III CC			Mean Square	F Value	P > F	
		00					
6	68.99904	4		11.49984	13.28	0.0001	
2	97.31047	7		48.65523	56.18	0.0001	
ng	Mean	N		Formulation	15		
	54.2533	3		5% dextrin			
	51.6700	3		3% dextrin			
	50.9720	3		control			
	50.5833	3		1% tapioca star	ch		
	50.2367	3		1% dextrin			
	49.3033	3		3% tapioca star	ch		
	48.0200	3		5% tapioca star	ch		
) vatic DF 12 20 DF 5 2 ng	Levels 7) 3 vations in data set DF Sum of Squares 3 166.3093 12 10.39342 20 176.7029 DF Type III 5 68.99904 2 97.31047 ng Mean 54.2533 51.6700 50.9720 50.5833 50.2367 49.3033 48.0200	Levels 7 7 7 9) 3 vations in data set = DF Sum of Squares 3 166.30951 12 10.39342 20 176.70293 DF Type III SS 5 68.99904 2 97.31047 ng Mean N 54.2533 3 51.6700 3 50.9720 3 50.2367 3 49.3033 3 48.0200 3	LevelsValue7contended1% s72, 3,032, 3,vations in data set = 21DFSum of Squares3166.309511210.3934220176.70293DFType III SS568.99904297.31047ngMeanN54.2533351.6700350.5833350.2367349.3033348.02003	LevelsValues7control, 1% dextrin, 1% starch, 3% starch)32, 3, 4vations in data set = 21 21 DFSum of SquaresMean Square3166.3095120.788681210.393420.8661120176.70293 11.49984 2097.3104748.65523ngMeanN54.253335% dextrin50.97203control50.583331% tapioca star50.236731% dextrin48.020035% tapioca star	Levels Values 7 control, 1% dextrin, 3% dextrin, 5 1% starch, 3% starch, 5% starch 1% starch, 3% starch, 5% starch) 3 2, 3, 4 vations in data set = 21 Mean Square DF Sum of Squares 3 166.30951 20.78868 24.00 12 10.39342 0.86611 20 20 176.70293 0.86611 20 DF Type III SS Mean Square F Value 5 68.99904 11.49984 13.28 20 97.31047 48.65523 56.18 Formulations 54.2533 3 5% dextrin 50.9720 3 control 50.5833 50.5833 3 1% tapioca starch 50.2367 3 1% dextrin 49.3033 3 5% tapioca starch	

Table B.11 ANOVA and Duncan's Multiple Range Test Table for coating pick-upof fried samples with different gum types.

0	lass		Levels	Va	lues				
F	ormulations		5	con	control, HPMC, xanthan gum, guar gum,				
				gua	guar-xanthan gum combination				
Frying time (min)		in)	3	2, 3	3, 4				
N	Sumber of obse	ervatio	ons in data s	et = 21					
		Sum of							
	Source	DF	Squares		Mean Square	F Value	$P_r > F$		
	Model	4	1835 134	420	458 78355	69 49	0.0001		
	Error	т 10	66 0220	τ <u>2</u> υ Λ	6 60220	07.77	0.0001		
		10	00.02300	J 	0.00230				
	Total	14	1901.15	720					
	0		T	1 00					
	Source	DF	I ype II	188	Mean Square	F Value	$\mathbf{h}^{\mathrm{L}} > \mathbf{h}$		
	Formulations	4	1835.13	3420	458.78355	69.49	0.0001		
Γ)uncan Group	oing	Mean	Ν	Formulation	IS			
	А		83.710	3	guar-xanthan g	um combina	tion		
	В		74.640	3	guar gum				
	В		73.300	3	xanthan gum				
	В		72.660	3	HPMC				
	С		50.290	3	control				

Table B.12 ANOVA and Duncan's Multiple Range Test Table for moisturecontent of fried samples with different gum types.

Class	Levels		Values						
Formulations	5		control, HPMC, xanthan gum, guar gum,						
		guar-xanthan gum combination							
Frying time (min)		3 2		2, 3, 4					
Number of observations in data set $= 15$									
		Sum of							
Source	DF	Squares			Mean Square	F Value	$P_r > F$		
Model	6	379.885	379.88572		63.31428	130.48	0.0001		
Error	8	3.88206			0.48525				
Total	14	383.76779							
Source	DF	Type III SS			Mean Square	F Value	$P_r > F$		
Batter Typ	e 4	228.845	228.84585		57.21146	117.90	0.0001		
Frying time	e 2	151.039	151.03987		75.51993	155.63	0.0001		
Duncan Gro	Mean	Ν	Formulations						
А		57.1667	3	guar-xanthan gum combination		ation			
А		56.0150	3		guar gum				
В		53.9450	3		xanthan gum				
В		53.4000	3		HPMC				
С		45.9833	3		control				

Table B.13 ANOVA and Duncan's Multiple Range Test Table for oil content of fried samples with different gum types.

Class			Levels		Values					
Formulations			5		control, HPMC, xanthan gum, guar gum,					
		guar-xanthan gum combination								
F	rying time (m	in)	3 2		2, 3,	2, 3, 4				
Number of observations in data set $= 15$										
			Sum of	•						
	Source DF Squares		5	Mean Square		F Value	$P_r > F$			
	Model	6	170.66274			28.44379	92.13	0.0001		
	Error	8	2.46978			0.30872				
	Total	14	173.13253							
-										
	Source DF Type III SS		I SS		Mean Square	F Value	$P_r > F$			
	Batter Type	4	137.682	233		34.42058	111.49	0.0001		
	Frying time	2	32.98041			16.49020	53.41	0.0001		
Duncan Grouping			Mean	N	[Formulations				
А		14.9400	3		control					
В		13.1300 3			xanthan gum					
	С		11.2367	3		НРМС				
	D		8.9933	3		guar gum				
E		6.3333	3.3333 3		guar-xanthan gum combination					

Table B.14 ANOVA and Duncan's Multiple Range Test Table for frying yield ofsamples with different gum types.

Class			Levels		Values				
Formulations			5		control, HPMC, xanthan gum, guar gum,				
		guar-xanthan gum combination							
F	rying time (m	in)	3		2, 3, 4				
,									
Number of observations in data set $= 15$									
	Cauraa	DE	Sum of			Maan Squara	E Value	$P_r > F$	
	Source	DF	Squares		Mean Square	F value			
-	Model	6	408.534	485		68.08914	46.00	0.0001	
	Error	8	11.8423	38		1.48029			
	Total	14	420.37724						
_									
	Source	DF	Type II	I SS	5	Mean Square	F Value	$P_r > F$	
-	Batter Type	4	179.13337			44.78334	30.25	0.0001	
	Frying time	2	229.401	229.40148		114.70074	77.48	0.0001	
Duncan Grouping			Mean	N	1	Formulations			
А		84.0467	84.0467 3		guar-xanthan gum combination				
AB		82.5433	3	i	xanthan g	gum			
BC		80.5967	3	1	НРМС				
С		79.7000	3)	guar gum				
	D		73.9533	3)	control			

Table B.15 ANOVA and Duncan's Multiple Range Test Table for fracturability offried samples with different gum types.

Class			Levels		Values				
Formulations			5		control, HPMC, xanthan gum, guar gum,				
			guar-xanthan gum combination						
F	Trying time (m	in)	3		2, 3, 4				
	·	,							
Number of observations in data set = 15									
			Sum of	•					
	Source	DF	Samoras		Mean Square	F Value	$P_r > F$		
					0.0000000	25.44	0.0001		
	Model	0	0.00000048		•	0.0000008	23.44	0.0001	
	Error	8	0.00000003			0.000000003			
	Total	14							
	Source	DF	Type II	ISS		Mean Square	F Value	$P_r > F$	
	Batter	4	0.00000027 0.00000021		7 0.00000067	0 00000067	21.35	0.0001	
	Туре					21.55	0.0001		
	Frying time	2				0.000000105	33.63	0.0001	
Duncan Grouping		Mean	Ν	N Formulations					
	А		0.00554	3		guar gum			
AB		0.00547	3		xanthan g	gum			
AB		0.00544	3		HPMC				
В		0.00541	3		guar-xanthan g	um combina	tion		
С		0.00514	3		control				
Table B.16 ANOVA and Duncan's Multiple Range Test Table for bulk density of

 fried samples with different gum types.

Class		Levels		Values					
F	ormulations		5		cont	ontrol, HPMC, xanthan gum, guar gum,			
					guar	-xanthan gum co	ombination		
F	rying time (m	in)	3		2, 3,	4			
N	umber of obs	ervati	ons in data se	et =	15				
		Sum of							
	Source	DF	Squares			Mean Square	F Value	$P_r > F$	
	Model	6	0.12833			0.02138	151.22	0.0001	
	Error	8	0.00113			0.00014			
	Total	14	0.12946						
-									
	Source	DF	Type III	SS		Mean Square	F Value	$P_r > F$	
-	Batter Type	4	0.11391			0.02847	210.34	0.0001	
	Frying time	2	0.01441			0.00720	50.97	0.0001	
D	uncan Grouj	ping	Mean	Ν		Formulatio	ns		
	А		1.10333	3		control			
	В		0.93233	3		guar gun	n		
В		0.92166	3		НРМС				
С		0.87566	3	3 xanthan gum					
С		0.85766	3	guar-xanthan gum combination					

Table B.17 ANOVA and Duncan's Multiple Range Test Table for porosity offried samples with different gum types.

C	lass	Levels		Values						
Fo	ormulations		5		cont	control, HPMC, xanthan gum, guar gum,				
				Ę	guar-	xanthan gum co	ombination			
Fr	Frying time (min)		3		2 3 4					
56 ()										
N	umber of obse	ervatio	ons in data se	et =	15					
1,1		or vari			10					
			Sum of							
	Source	DF	Sauaraa			Mean Square	F Value	$P_r > F$		
-	NC 11		Squares			0.0000	00.15	0.0001		
	Model	6	0.05957			0.00992	80.15	0.0001		
	Error	8	0.00099			0.00012				
	Total	14	0.06056							
	Source	DF	Type III	SS		Mean Square	F Value	$P_r > F$		
	Batter Type	4	0.05536			0.01384	111.74	0.0001		
	Frying time	2	0.00420			0.0021	16.98	0.0001		
D	uncan Group	oing	Mean	N		Formulation	18			
	А		0.21653	3		xanthan gum				
	В		0.19320	3		guar-xan	than	gum		
					combination					
	С		0.12765	3		HPMC				
	С		0.10670 3			guar gum				
	D		0.04769	3		control				

Table B.18 ANOVA and Duncan's Multiple Range Test Table for Hunter L valueof fried samples with different gum types.

Class		Levels		Values					
Formulations		5		cont	control, HPMC, xanthan gum, guar gum,				
				guar-xanthan gum combination					
Frying time (min	1)	3 2,		2, 3,	4				
Number of obser	rvatio	ons in data	set =	= 15					
Source DE		Sum of	f		Moon Squara	E Value	D > E		
Source	DF	Square	s		Mean Square	r value	$P_{\rm r} > \Gamma$		
Model	6	205.12	205.12676		34.18779	18.71	0.0001		
Error	8	14.619	14.61981		1.82747				
Total	14	219.74657							
Source	DF	Type I	II SS		Mean Square	F Value	$P_r > F$		
Batter Type	4	120.36	730		30.09182	16.47	0.0001		
Frying time	2	84.759	45		42.37972	23.19	0.0001		
Duncan Groupi	ing	Mean	N	-	Formulation	18			
А		61.043	3		guar-xanthan g	um combina	tion		
AB		59.163	3		xanthan g	gum			
BC		57.930	3		guar gum				
C 56.		56.217	3		HPMC				
D 52.703 3		3		control					

Table B.19 ANOVA and Duncan's Multiple Range Test Table for Hunter a valueof fried samples with different gum types.

С	lass		Levels		Values					
F	ormulations		5		cont	control, HPMC, xanthan gum, guar gum,				
					guar	-xanthan gum co	ombination			
F	rying time (m	in)	3 2,		2, 3,	4				
Number of observation			ons in data	set =	= 15					
		Sum of	f							
	Source	DF	Square	s		Mean Square	F Value	$P_r > F$		
-	Model	6	45.41208			7.56868	11.46	0.0015		
	Error	8	5.2828	5.28289		0.66036				
	Total	14	50.69497							
-										
	Source	DF	Type I	II SS		Mean Square	F Value	$P_r > F$		
-	Batter Type	4	21.981	10		5.49527	8.32	0.0060		
	Frying time	2	23.430	97		11.71548	17.74	0.0011		
D	uncan Grouj	ping	Mean	Ν		Formulation	18			
	А		9.2200	3		control				
	B 6.8933		3		HPMC					
В		6.8433	3		guar gum					
B 6.043		6.0433	3		xanthan gum					
	В		5.7933	3		guar-xanthan gum combination				

Table B.20 ANOVA and Duncan's Multiple Range Test Table for ΔE of fried samples with different gum types.

Class		Levels		Values				
Formulations		5		control, HPMC, xanthan gum, guar gum,				
			Ę	guar-xanthan gum	combination			
Frying time (r	nin)	3 2, 2		2, 3, 4				
Number of ob	servati	ons in data s	set =	15				
Carrier DE		Sum of		Moon Squara	E Valua	D \ F		
Source	DF	Squares	3	wiean Square	r value	Г _Г ~ Г		
Model	6	192.763	376	32.12729	27.66	0.0001		
Error	8	9.29325	5	1.16165				
Total	14	202.057	702					
Source	DF	Type II	I SS	Mean Square	F Value	$P_r > F$		
Batter Type	; 4	130.045	506	32.51126	27.99	0.0001		
Frying time	2	62.7187	70	31.35935	27.00	0.0001		
Duncan Grou	ıping	Mean	Ν	Formulati	ons			
А		50.9720	3	control				
В		47.2500	3	guar gu	ım			
С	С		3	xanthan gum				
C 43.6500 3		3	HPMC	НРМС				
С		43.0867	3	guar-xanthan	guar-xanthan gum combination			

B.21. ANOVA and Duncan's Multiple Range Test Table for coating pick up of fried samples with different starch or gum types.

Class		Levels		Values			
Formulations		11		ntrol, 1% dextrin, 3% dextrin, 5% dextrin,			
				1% starch, 3% starch, 5% starch			
]	HPMC, xanthan gum, guar gum,			
			Į	guar-xanthan gum combination			
Frying time (min)		3		2, 3, 4			
Number of ob	servatio	ons in data	set =	33			
		Sum of	f				
Source	DF	Square	S	Mean Square F value $P_r > F$			
Model	10	4550.6	4123	455.064123 107.78 0.0001			
Error	22	80.220	10	3.64636			
Total	32	4630.8	6133				
Source	DF	Туре	III SS	S Mean Square F Value $P_r > F$			
Formulation	ns 10	4550.	64123	3 455.064123 107.78 0.0001			
Duncan Grou	iping	Mean	N	Formulations			
А		83.710	3	guar-xanthan gum combination			
В		75.650	3	5% starch			
В		74.645	2	guar gum			
В		73.300	3	xanthan gum			
В		72.660	3	HPMC			
С		64.120	2	3% starch			
D		51.940	3	5% dextrin			
D		51.840	3	3% dextrin			
D		51.505	2	1% starch			
D		50.920	3	1% dextrin			
D		50.290	3	control			

B.22. ANOVA and Duncan's Multiple Range Test Table for moisture content of fried samples with different starch or gum types.

Class Levels		Values							
Formulations		11		con	trol, 1% dextrin,	3% dextrin,	5% dextrin,		
				1% starch, 3% starch, 5% starch					
				HPMC, xanthan gum, guar gum,					
				guar	-xanthan gum co	mbination			
Frying time (min)		3		2, 3,	, 4				
Number of obser	ons in data	set =	= 33						
		Sum o	f						
Source	DF	Square	es		Mean Square	F Value	$P_r > F$		
Model	10	450.14	330		45.0143	22.68	0.0001		
Error	22	43.665	20		1.98478				
Total	32	493.80	850						
Source	DF	Type	III S	S	Mean Square	F Value	$P_r > F$		
Source	DI	rype	mo						
Formulations	10	450.1	4330)	45.014330	22.68	0.0001		
Formulations Duncan Groupi	10 ng	450.1 Mean	4330 N)) N	45.014330 Formulation	22.68	0.0001		
Formulations Duncan Groupi A	10 ng	450.1 Mean 57.900	4330 N 3)) N 3	45.014330 Formulation guar-xanthan g	22.68 Is um combina	0.0001		
Formulations Duncan Groupi A AB	10 ng	450.1 Mean 57.900 56.370	4330 4330 N 3 3)) } }	45.014330 Formulation guar-xanthan g guar gum	22.68 Is um combina	0.0001		
Formulations Duncan Groupi A AB BC	10 ng	450.1 Mean 57.900 56.370 54.955	111 3 433(N 3 3 3) N 3 3	45.014330 Formulation guar-xanthan g guar gum 5% starch	22.68 Is um combina	0.0001		
Formulations Duncan Groupi A AB BC BCD	10 ng	450.1 Mean 57.900 56.370 54.955 54.190	111 3 4330 N 3 3 3 3) } } }	45.014330 Formulation guar-xanthan g guar gum 5% starch HPMC	22.68 Is um combina	0.0001		
Formulations Duncan Groupi A AB BC BCD BCD	10 ng	450.1 Mean 57.900 56.370 54.955 54.190 53.840	433(433(N 3 3 3 3 3 3 3)) } } }	45.014330 Formulation guar-xanthan g guar gum 5% starch HPMC 3% starch	22.68 Is um combina	0.0001		
Formulations Duncan Groupi A AB BC BCD BCD CD	10 ng	450.1 Mean 57.900 56.370 54.955 54.190 53.840 53.135	111 3 4330 3 3 3 3 3 3 3 3 3 3 3 3 3)) } } } }	45.014330 Formulation guar-xanthan g guar gum 5% starch HPMC 3% starch xanthan gum	22.68 Is um combina	0.0001		
Formulations Duncan Groupi A AB BC BCD BCD CD D	10 ng	450.1 Mean 57.900 56.370 54.955 54.190 53.840 53.135 52.310	433(433(N 3 3 3 3 3 3 3 3 3 3 3 3 3)) 3 3 3 3 3 3 3 3	45.014330 Formulation guar-xanthan g guar gum 5% starch HPMC 3% starch xanthan gum 1% dextrin	22.68 Is um combina	0.0001		
Formulations Duncan Groupi A AB BC BCD BCD CD D D	10 ng	450.1 Mean 57.900 56.370 54.955 54.190 53.840 53.135 52.310 52.060	4330 4330 N 3 3 3 3 3 3 3 3 3 3 3 3 3)) 1 3 3 3 3 3 3 3 3 3 3	45.014330 Formulation guar-xanthan g guar gum 5% starch HPMC 3% starch xanthan gum 1% dextrin 1% starch	22.68 Is um combina	0.0001		
Formulations Duncan Groupi A AB BC BCD BCD CD D D E	10 ng	450.1 Mean 57.900 56.370 54.955 54.190 53.840 53.135 52.310 52.060 48.680	4330 4330 N 3 3 3 3 3 3 3 3 3 3 3 3 3)) 1 3 3 3 3 3 3 3 3 3 3	45.014330 Formulation guar-xanthan g guar gum 5% starch HPMC 3% starch xanthan gum 1% dextrin 1% starch 3% dextrin	22.68 Is um combina	0.0001		
Formulations Duncan Groupi A AB BC BCD BCD CD D D E F	10 ng	450.1 Mean 57.900 56.370 54.955 54.190 53.840 53.135 52.310 52.060 48.680 46.250	4330 4330 N 3 3 3 3 3 3 3 3 3 3 3 3 3) N 3 3 3 3 3 3 3 3 3 3 3 3 3	45.014330 Formulation guar-xanthan g guar gum 5% starch HPMC 3% starch xanthan gum 1% dextrin 1% starch 3% dextrin control	22.68 Is um combina	0.0001		

B.23. ANOVA and Duncan's Multiple Range Test Table for oil content of fried samples with different starch or gum types.

Class		Levels		Values					
Formulations		11		control, 1% dextrin, 3% dextrin, 5% dextrin,					
				1% s	tarch, 3% starch	, 5% starch			
				HPM	IC, xanthan gum	, guar gum,			
				guar-xanthan gum combination					
Frying time (min)		3		2, 3,	4				
Number of observ	vatio	ons in data s	et =	= 33					
		Sum of			M G		DND		
Source L	JF	Squares			Mean Square	F value	$P_r > F$		
Model 1	0	157.422	53		15.742253	17.27	0.0001		
Error 2	22	11.8514	0		0.5387				
Total 3	32	169.273	93						
Source	DF	Type II	IS	S	Mean Square	F Value	$P_r > F$		
Formulations	10	157.42253		5	15.742253	17.27	0.0001		
Duncan Groupin	ng	Mean	N	ſ	Formulation	15			
А		15.2500	4		control				
А		14.8500	2		5% dextr	in			
AB		14.2000	2		3% dextrin				
BC		12.3800	2		xanthan gum				
BC		12.3300	2		1% dextrin				
С		11.8100	2		HPMC				
С		11.6600	2	1 4	1% starch				
DC		11.4400	2		3% starch				
DC		10.4400	2		5% starch				
D		9.4900	2	,	guar gum				
Е		6.0600	2		guar-xanthan g	um combina	tion		

B.24. ANOVA and Duncan's Multiple Range Test Table for fracturability of fried samples with different starch or gum types.

Class		Levels	V	Values					
Formulations		11	C	control, 1% dextrin,	3% dextrin,	5% dextrin,			
				1% starch, 3% starch, 5% starch					
		Н	HPMC, xanthan gum, guar gum,						
		g	guar-xanthan gum combination						
Frying time (m	in)	3	2	2, 3, 4					
Number of obs	ervatio	ons in data se	et = 3	33					
0	DE	Sum of							
Source	DF	Squares		Mean Square	F Value	$P_r > F$			
Model	10	0.000000	55	0.000000055	3.55	0.0063			
Error	22	0.000000	34	0.000000015					
Total	32	0.000000	90						
Source	DF	Type III	SS	Mean Square	F Value	$P_r > F$			
Formulations	s 10	0.00000	055	0.00000055	3.55	0.0063			
Duncan Grouj	ping	Mean	N	Formulation	18				
А		0.0055900	3	guar gum					
AB		0.0055290	3	xanthan g	gum				
ABC		0.0054750	3	HPMC					
ABCD		0.0054200	3	guar-xanthan g	um combina	tion			
ABCD		0.0054140	3	5% starch					
BCDE		0.0053380	3	5% dextrin					
BCDE		0.0053270	3	3% starch					
BCDE		0.0052970	3	1% starch					
CDE		0.0052510	3	3% dextrin					
DE		0.0051990	3	1% dextrin					
Е		0.0051590	3	s control					

B.25. ANOVA and Duncan's Multiple Range Test Table for bulk density of fried samples with different starch or gum types.

Class	Levels	Val	lues		
Formulations	11	con	trol, 1% dextrin,	3% dextrin,	5% dextrin,
		1%	starch, 3% starch	, 5% starch	
		HPN	MC, xanthan gum	n, guar gum,	
		guar	r-xanthan gum co	ombination	
Frying time (min)	3	2, 3	, 4		
Number of observati	ons in data se	t = 33			
Second DE	Sum of		Maan Carrena	E Malara	DNE
Source DF	Squares		Mean Square	F value	$P_r > F$
Model 10	0.19811		0.019811	20.78	0.0001
Error 22	0.02097		0.000953		
Total 32	0.21909				
Source DI	Type II	I SS	Mean Square	F Value	$P_r > F$
Formulations 10	0.19811		0.019811	20.78	0.0001
Duncan Grouping	Mean	N	Formulation	18	
А	1.09000	3	control		
А	1.07000	3	1% dextr	in	
А	1.06200	3	3% dextrin		
А	1.03400	3	5% dextrin		
В	0.97700	3	1% starch		
BC	0.96700	3	3% starch		
BC	0.94600	3	5% starch		
BC	0.93200	3	guar gum		
CD	0.91800	3	HPMC		
DE	0.87200	3	xanthan gum		
Е	0.84800	3	guar-xanthan g	um combina	tion

B.26. ANOVA and Duncan's Multiple Range Test Table for Hunter L value of fried samples with different starch or gum types.

Class		Levels		Val	ues			
Formulations		11		cont	trol, 1% dextrin,	3% dextrin,	5% dextrin,	
				1% s	starch, 3% starch	, 5% starch		
				HPMC, xanthan gum, guar gum,				
				guar	-xanthan gum co	mbination		
Frying time (min)		3		2, 3,	4			
Number of obse	ervatio	ons in data	set =	= 33				
		Sum of	f					
Source	DF	Square	s		Mean Square	F value	$P_r > F$	
Model	10	814.81	985		81.48198	16.93	0.0001	
Error	22	158.82	800		7.21945			
Total	32	973.64	785					
Source	DF	Type	III S	S	Mean Square	F Value	$P_r > F$	
Formulations	10	814.8	1985	5	81.481985	16.93	0.0001	
Duncan Group	oing	Mean	Ν	I	Formulations			
А		63.010	4	ļ	guar-xanthan g	um combina	tion	
В		59.760	4	ļ	xanthan g	gum		
В		58.480	4	ļ	guar gum			
В		57.620	4		НРМС			
В		57.140	4	ŀ	5% starch			
В		56.400	4	ļ	3% starch			
С		52.600	4	ļ	1% starch			
CD		52.250	4	ļ	control			
CD		51.200	4	ļ	1% dextrin			
CD		49.820	4	Ļ	3% dextrin			
D		48.900	4	Ļ	5% dextrin			

B.27. ANOVA and Duncan's Multiple Range Test Table for Hunter a value of fried samples with different starch or gum types.

Class			Levels		Valı	ues			
F	ormulations		11		cont	control, 1% dextrin, 3% dextrin, 5% de			
					1% s	starch, 3% starch	, 5% starch		
					HPM	IC, xanthan gum	, guar gum,		
					guar	-xanthan gum co	mbination		
Frying time (min)			3		2, 3,	4			
N	lumber of obse	ervatio	ons in data s	et =	= 33				
			Sum of					D. D	
	Source	DF	Squares			Mean Square	F Value	$P_r > F$	
	Model	10	506.425	81		50.64258	36.09	0.0001	
	Error	22	46.3112	0		2.10505			
	Total	32	552.737	01					
-									
	Source	DF	Type I	II S	S	Mean Square	F Value	$P_r > F$	
	Formulations	10	506.42	581		50.64258	36.09	0.0001	
D	uncan Group	oing	Mean	N	I	Formulation	15		
	А		15.0800	4	ļ	5% dextrin			
	А		14.9000	4	ļ	3% dextr	in		
	В		12.6400	4	ļ	1% dextrin			
	С		9.9200	4	ļ	control			
	CD		8.4400	4	ļ	1% starch			
	DE		7.8400	4	ļ	3% starch			
	DEF		7.1400	4	ļ	5% starch			
	EF		6.5500	4	ļ	НРМС			
	EF 6		6.4000	4	ļ	guar gum			
	F		5.7800	4	ļ	xanthan gum			
	F		5.4500	4	ļ	guar-xanthan gum combination			

APPENDIX C

WATER BINDING CAPACITY

Table C.1 Water binding capacities (WBC) of different starches and gums(Altunakar, 2003).

	WBC (w/w)	
Corn and wheat	1.04	
Pre-gelatinized tapioca starch	4.80	
HPMC	12.10	
Xanthan gum	11.49	
Guar gum	14.36	

APPENDIX D

FIGURES OF DEEP-FAT FRIED CARROT SLICES



Figure D.1 Carrot and control batter fried for 2 & 4 minutes.



Figure D.2 Carrot and 1% dextrin added batter fried for 2 & 4 minutes.



Figure D.3 Carrot and 5% dextrin added batter fried for 2 & 4 minutes.



Figure D.4 Carrot and 1% pre-gelatinized tapioca starch added batter fried for 2 & 4 minutes.



Figure D.5 Carrot and 5% pre-gelatinized tapioca starch added batter fried for 2 & 4 minutes.



Figure D.6 Carrot and xanthan gum added batter fried for 2 & 4 minutes.



Figure D.7 Carrot and HPMC added batter fried for 2 & 4 minutes.



Figure D.8 Carrot and guar-xanthan gum combination added batter fried for 2 & 4 minutes.

APPENDIX E

SURFACE PLOTS OF IMAGE PROCESSING



Figure E 1: Surface plot of control coated carrot



Figure E 2: Surface plot of guar-xanthan gum combination added batter coated carrot.